Name _____ Class _____ Date _____

Worksheet 6.1 *continued*

An *ionic bond* is formed when one or more electrons are transferred from one atom to another. Ionic compounds usually form between metals and nonmetals.

Use the information in the box to help you write the correct formula for each ionic compound.

Al	
Br	
$Mg(NO_3)_2$	
Li_2CO_3	
$(NH_4)_2Cr_2O_7$	

Aluminum (Al) ion	**+3**
Bromide (Br) ion	**−1**
Dichromate (Cr_2O_7) ion	**−2**
Chromium (Cr) atom	**0**
Chromium (Cr) ion	**+3**
Lithium (Li) ion	**+1**
Magnesium (Mg) ion	**+2**
Nitrate (NO_3) ion	**−1**
Oxygen (O) atom	**0**
Oxygen (O) ion	**−2**
Zinc (Zn) ion	**+2**

Each word below has a Column A, and the letter of the best match will then be written in the blank to the left of Column B.

Column A	Column B
_____ **106.** An endothermic reaction	**a.** reactants
_____ **107.** An exothermic reaction	**b.** energy
_____ **108.** Energy is conserved	**c.** products
_____ **109.** Substances that are already combined chemically	**d.** endothermic reaction
_____ **110.** Energy is released	**e.** exothermic reaction

Use the figure to the right to answer the questions below. Write the letter of the correct answer in the blank provided.

_____ **111.** Which area represents the reactants?

_____ **112.** Which area represents the products?

_____ **113.** Energy is absorbed during the reaction.

_____ **114.** Energy is released during the reaction.

_____ **115.** Energy is conserved during the reaction.

Study Guide For Contrnt Mastery

Christian G. Meyer

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Table of Contents Study Guide For Contrnt Mastery

1. Understanding the eBook Study Guide For Contrnt Mastery
    - The Rise of Digital Reading Study Guide For Contrnt Mastery
    - Advantages of eBooks Over Traditional Books
2. Identifying Study Guide For Contrnt Mastery
    - Exploring Different Genres
    - Considering Fiction vs. Non-Fiction
    - Determining Your Reading Goals
3. Choosing the Right eBook Platform
    - Popular eBook Platforms
    - Features to Look for in an Study Guide For Contrnt Mastery
    - User-Friendly Interface
4. Exploring eBook Recommendations from Study Guide For Contrnt Mastery
    - Personalized Recommendations

- Study Guide For Contrnt Mastery User Reviews and Ratings
- Study Guide For Contrnt Mastery and Bestseller Lists

5. Accessing Study Guide For Contrnt Mastery Free and Paid eBooks
   - Study Guide For Contrnt Mastery Public Domain eBooks
   - Study Guide For Contrnt Mastery eBook Subscription Services
   - Study Guide For Contrnt Mastery Budget-Friendly Options
6. Navigating Study Guide For Contrnt Mastery eBook Formats
   - ePub, PDF, MOBI, and More
   - Study Guide For Contrnt Mastery Compatibility with Devices
   - Study Guide For Contrnt Mastery Enhanced eBook Features
7. Enhancing Your Reading Experience
   - Adjustable Fonts and Text Sizes of Study Guide For Contrnt Mastery
   - Highlighting and Note-Taking Study Guide For Contrnt Mastery
   - Interactive Elements Study Guide For Contrnt Mastery
8. Staying Engaged with Study Guide For Contrnt Mastery
   - Joining Online Reading Communities
   - Participating in Virtual Book Clubs
   - Following Authors and Publishers Study Guide For Contrnt Mastery
9. Balancing eBooks and Physical Books Study Guide For Contrnt Mastery
   - Benefits of a Digital Library
   - Creating a Diverse Reading Collection Study Guide For Contrnt Mastery
10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
11. Cultivating a Reading Routine Study Guide For Contrnt Mastery
    - Setting Reading Goals Study Guide For Contrnt Mastery
    - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Study Guide For Contrnt Mastery
    - Fact-Checking eBook Content of Study Guide For Contrnt Mastery

- Distinguishing Credible Sources

13. Promoting Lifelong Learning
    - Utilizing eBooks for Skill Development
    - Exploring Educational eBooks
14. Embracing eBook Trends
    - Integration of Multimedia Elements
    - Interactive and Gamified eBooks

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