

Name Mel

Locker # \_\_\_\_\_

**Bonding & Molecular Structure****LEWIS STRUCTURES**

Indicate the total # of VALENCE electrons for each molecule (VE's). Calculate the number of bonded electrons from the skeleton (BE's). Calculate the Remaining Electrons (RE's = VE's - BE's) and distribute the remaining electrons onto the skeletal structure. Finally, satisfy the octets by creating double and triple bonds. Your final structures should be given below and your scratch paper should clearly show your work.

$F_2$ VE's = <u>14</u> BE's = <u>2</u> RE's = <u>12</u> $:\ddot{F}-\ddot{F}:$ Shape: <u>linear</u>	$H_2$ VE's = <u>2</u> BE's = <u>2</u> RE's = <u>0</u> $H-H$ Shape: <u>linear</u>	$HF$ VE's = <u>8</u> BE's = <u>2</u> RE's = <u>6</u> $H-\ddot{F}:$ Shape: <u>linear</u>	$NH_3$ VE's = <u>8</u> BE's = <u>6</u> RE's = <u>2</u> $H-\overset{\cdot\cdot}{\underset{\cdot\cdot}{N}}-H$ Shape: <u>trigonal pyramidal</u>
$CH_4$ VE's = <u>8</u> BE's = <u>8</u> RE's = <u>0</u> $H-\overset{\cdot\cdot}{\underset{\cdot\cdot}{C}}-H$ Shape: <u>tetrahedral</u>	$C_2H_6$ VE's = <u>14</u> BE's = <u>14</u> RE's = <u>0</u> $H-\overset{\cdot\cdot}{\underset{\cdot\cdot}{C}}-\overset{\cdot\cdot}{\underset{\cdot\cdot}{C}}-H$ Shape: <u>tetrahedral (both)</u>	$AlH_3$ VE's = <u>6</u> BE's = <u>6</u> RE's = <u>0</u> $H-Al-H$ Shape: <u>trigonal planar</u>	$BH_3$ VE's = <u>6</u> BE's = <u>6</u> RE's = <u>0</u> $H-\overset{\cdot\cdot}{\underset{\cdot\cdot}{B}}-H$ Shape: <u>trigonal planar</u>
$C_2H_4$ VE's = <u>12</u> BE's = <u>12</u> RE's = <u>0</u> $H-\overset{\cdot\cdot}{\underset{\cdot\cdot}{C}}=\overset{\cdot\cdot}{\underset{\cdot\cdot}{C}}-H$ Shape around one C: <u>trigonal planar</u>	$C_2F_4$ VE's = <u>36</u> BE's = <u>12</u> RE's = <u>24</u> $:\ddot{F}-\overset{\cdot\cdot}{\underset{\cdot\cdot}{C}}=\overset{\cdot\cdot}{\underset{\cdot\cdot}{C}}-\ddot{F}:$ Shape around one C: <u>trigonal planar</u>	$CO$ VE's = <u>10</u> BE's = <u>6</u> RE's = <u>4</u> $\overset{\cdot\cdot}{C}\equiv\overset{\cdot\cdot}{O}$ Shape: <u>linear</u>	$O_2$ VE's = <u>12</u> BE's = <u>4</u> RE's = <u>8</u> $:\overset{\cdot\cdot}{O}=\overset{\cdot\cdot}{O}:$ Shape: <u>linear</u>

# South Pasadena Bonding And Molecular Structure Worksheet Answers

**Michael Brown**



**South Pasadena Bonding And Molecular Structure Worksheet Answers:**

## Embracing the Track of Phrase: An Mental Symphony within **South Pasadena Bonding And Molecular Structure Worksheet Answers**

In a global eaten by screens and the ceaseless chatter of fast interaction, the melodic splendor and psychological symphony created by the prepared term frequently fade into the backdrop, eclipsed by the persistent noise and distractions that permeate our lives. However, nestled within the pages of **South Pasadena Bonding And Molecular Structure Worksheet Answers** a charming fictional prize full of raw thoughts, lies an immersive symphony waiting to be embraced. Constructed by a masterful composer of language, this charming masterpiece conducts readers on a psychological trip, skillfully unraveling the hidden tunes and profound impact resonating within each cautiously constructed phrase. Within the depths of the emotional evaluation, we will discover the book is main harmonies, analyze its enthralling writing type, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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