



**PHYSICAL SCIENCES (CHEMISTRY)
GRADE 11 TERM 3 FORMAL EXPERIMENT - PART 2**

TIME: 1 HOUR

**VERIFICATION OF BOYLE'S LAW
INDIVIDUAL WORK (should be done under strict supervision).**

1. Write a scientific report which must include the following:

1.1 Aim

(1)

1.2 Apparatus and method

(5)

2. Observations and Recording of results (Copy the table below onto your book or sheet of paper and transfer the results you recorded when conducting the experiment). Calculate pV for each set of results and record your answers in the table.

Volume (V) cm^3	Pressure (p) kPa		pV (J)

(6)

3. Analysis of the results:

3.1 Identify the following variables for the experiment:

- Independent
- Dependent
- Controlled

(3)

3.2 Use the graph paper provided by your teacher to draw graphs of V versus p and V versus $\frac{1}{p}$.

(18)

4. Conclusion

4.1 Examine the results in your table as well as the two graphs you have drawn, what happens to the volume of the gas in the tube as the pressure is decreased.

(1)

4.2 Write down a conclusion for this experiment.

(3)

5

Questions and discussion:

5.1 Explain why you have to wait for a few minutes before you can take readings.

(2)

5.2 Explain the relationship between pressure and volume in terms of the kinetic theory of gases.

(4)

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Lei Huang



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Table of Contents Term 3 Grade Physical Science Emplar Test 2014

1. Understanding the eBook Term 3 Grade Physical Science Emplar Test 2014
 - The Rise of Digital Reading Term 3 Grade Physical Science Emplar Test 2014
 - Advantages of eBooks Over Traditional Books
2. Identifying Term 3 Grade Physical Science Emplar Test 2014
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Term 3 Grade Physical Science Emplar Test 2014
 - User-Friendly Interface
4. Exploring eBook Recommendations from Term 3 Grade Physical Science Emplar Test 2014
 - Personalized Recommendations
 - Term 3 Grade Physical Science Emplar Test 2014 User Reviews and Ratings
 - Term 3 Grade Physical Science Emplar Test 2014 and Bestseller Lists
5. Accessing Term 3 Grade Physical Science Emplar Test 2014 Free and Paid eBooks

- Term 3 Grade Physical Science Emplar Test 2014 Public Domain eBooks
- Term 3 Grade Physical Science Emplar Test 2014 eBook Subscription Services
- Term 3 Grade Physical Science Emplar Test 2014 Budget-Friendly Options
- 6. Navigating Term 3 Grade Physical Science Emplar Test 2014 eBook Formats
 - ePub, PDF, MOBI, and More
 - Term 3 Grade Physical Science Emplar Test 2014 Compatibility with Devices
 - Term 3 Grade Physical Science Emplar Test 2014 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Term 3 Grade Physical Science Emplar Test 2014
 - Highlighting and Note-Taking Term 3 Grade Physical Science Emplar Test 2014
 - Interactive Elements Term 3 Grade Physical Science Emplar Test 2014
- 8. Staying Engaged with Term 3 Grade Physical Science Emplar Test 2014
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Term 3 Grade Physical Science Emplar Test 2014
- 9. Balancing eBooks and Physical Books Term 3 Grade Physical Science Emplar Test 2014
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Term 3 Grade Physical Science Emplar Test 2014
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Term 3 Grade Physical Science Emplar Test 2014
 - Setting Reading Goals Term 3 Grade Physical Science Emplar Test 2014
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Term 3 Grade Physical Science Emplar Test 2014
 - Fact-Checking eBook Content of Term 3 Grade Physical Science Emplar Test 2014
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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