

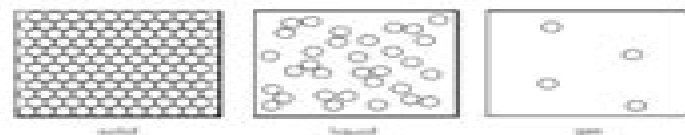
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Chapter 1

Exercise 1.1

- 1 A solid has a fixed **mass** and **volume** / **shape**.
A liquid has a fixed **mass** but its **shape** changes to that of the container in which it is placed.
A gas has no fixed **shape** or **volume**. A gas completely fills the container that it is in.

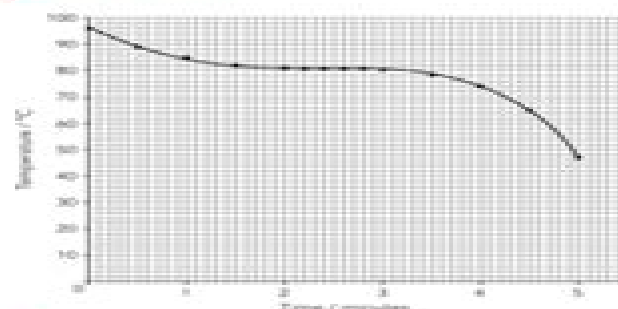
2



- 3 a B
b C
c A
d D
- 4 A freezing (solidification)
B melting
C condensation
D evaporation or boiling
- 5 a radon
b radon and nitrogen
c nitrogen
d cobalt
e The sample of ethanoic acid is impure. The presence of impurities raises the boiling point of a substance.

Exercise 1.2

6



- 7 The student would need to use an oil bath (in place of the water bath) so that the higher temperature could be reached.
- 8 The substance is freezing (solidifying) / turning from liquid to solid.
- 9 The temperature stays constant because energy is being released as the substance solidifies / the molecules are giving out heat as they stop moving from place to place and become organised in a structured lattice arrangement / in the solid the molecules can only vibrate about fixed points / the heat released by the formation of new interactions (forces) between the particles keeps the temperature constant until all the substance is solid.

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1. Understanding the eBook Temperature Controls Study Workbook Answers
 - The Rise of Digital Reading Temperature Controls Study Workbook Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Temperature Controls Study Workbook Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Temperature Controls Study Workbook Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Temperature Controls Study Workbook Answers

- Personalized Recommendations
 - Temperature Controls Study Workbook Answers User Reviews and Ratings
 - Temperature Controls Study Workbook Answers and Bestseller Lists
5. Accessing Temperature Controls Study Workbook Answers Free and Paid eBooks
- Temperature Controls Study Workbook Answers Public Domain eBooks
 - Temperature Controls Study Workbook Answers eBook Subscription Services
 - Temperature Controls Study Workbook Answers Budget-Friendly Options
6. Navigating Temperature Controls Study Workbook Answers eBook Formats
- ePub, PDF, MOBI, and More
 - Temperature Controls Study Workbook Answers Compatibility with Devices
 - Temperature Controls Study Workbook Answers Enhanced eBook Features
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 - Highlighting and Note-Taking Temperature Controls Study Workbook Answers
 - Interactive Elements Temperature Controls Study Workbook Answers
8. Staying Engaged with Temperature Controls Study Workbook Answers
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Temperature Controls Study Workbook Answers
9. Balancing eBooks and Physical Books Temperature Controls Study Workbook Answers
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Temperature Controls Study Workbook Answers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Temperature Controls Study Workbook Answers
- Setting Reading Goals Temperature Controls Study Workbook Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Temperature Controls Study Workbook Answers

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 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
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- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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