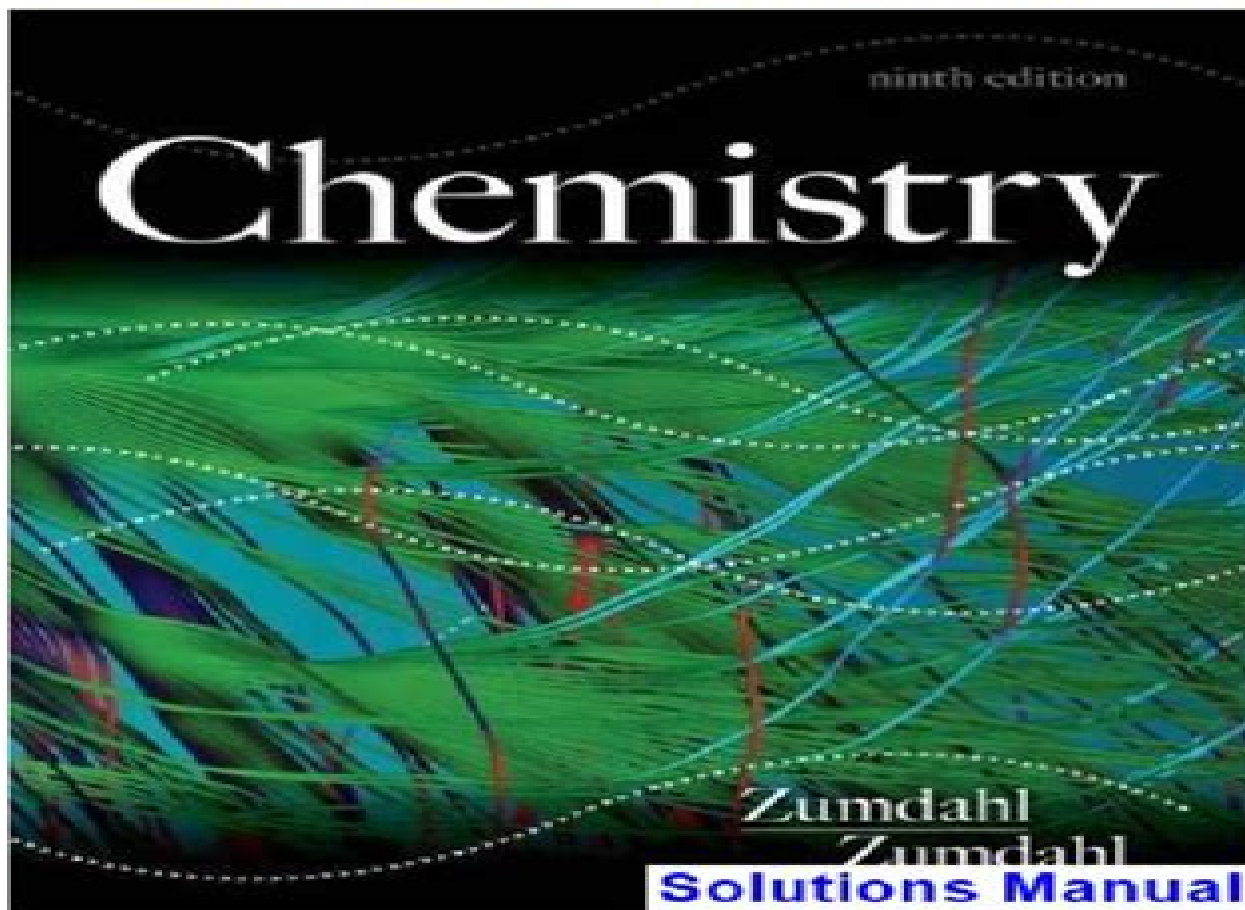


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 temperature 2 the beaker now contains sodium chloride and water and the

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web reacting two solutions eg acid and alkali place the polystyrene cup inside the glass beaker to make it more stable
 measure an appropriate volume of each liquid eg 25 cm³ place one of the

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 exothermic than with strong acids and bases for example one source which gives the enthalpy change of neutralization of
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 eg 1 hcl aq + naoh aq → nacl aq + h₂o l for which Δh = 57 kJ mol⁻¹ eg 2 h₂no₃ aq + koh aq → kno₃ aq + h₂o l for which Δh = 57 kJ mol⁻¹

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