AP Chemistry Practice Test Chapter 8 and 9

Multiple Choice

Identify the choice that best completes the statement or answers the question.

	13	Which pair of atoms should form the most polar bond	?
			and F
			and N
		c. F and O	
	23	Which pair of ions should form the ionic lattice with the highest energy?	
		a. Na ¹⁺ and Br ¹⁻ d. Li	i ¹ * and O ² *
		b. Li ^{1*} and F ¹⁻ e. K	1+ and F1-
		c. Cs1+ and F1-	
	3.	3. The electrostatic force of attraction is greatest in which	h compound?
		a. BaO d. M	lgS
		b. MgO e. C	aO
		c. CaS	
	4.	Which molecule has the weakest bond?	
		a, CO d. N	
		b. O ₂ e. C	l ₂
		c. NO	
	5.	5. For which species are octet resonance structures neces	
			eF ₂
		b. SO ₂ c. Si	O ₄ ²⁻
		c. CO ₂	
	6.	6. How are the bonding pairs arranged in the best Lewis	
			=0-O
		b. O=O-O e. O	=0=0
		c. O=O-O	
	7.	7. Which species has the shortest bond length?	
		a. CN ¹ - d. So	O ₃
		b. O ₂ e. C	0,
		e. SO ₂	
	8.	8. Which species requires the least amount of energy to	remove an electron from the outermost energy level?
	10.00	a. Na ^{1*} d. O	
			lg ² *
		c. F1-	
	9.	Which species has a valid non-octet Lewis structure?	
			eCl ₄
		b. SiF ₄ c. C	O,5-

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Rose Arny

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