

Skills Worksheet

Concept Review Chapter 5: Section 1

Section: Simple Ions

Complete each statement below by choosing a term from the following list. Terms may be used more than once.

10

11

ion

octet

12

13

anion

cation

1. An atom or group of atoms that has a positive or negative electric charge because it has lost or gained electrons is a(n) ion.
2. The sodium ion has 11 protons, 12 neutrons, and 10 electrons.
3. In most chemical reactions, atoms tend to match the outer electron configuration of the noble gases. This is called the octet rule.

Complete each statement below by writing the correct term or phrase.

4. An anion is an ion with a negative charge.
5. Valence electrons occupy the outermost energy level of an atom.
6. The chloride ion and the chlorine atom have different electron configurations.
7. A cation is an ion with a positive charge.
8. Many stable ions have an electron configuration of a noble gas.

Answer the following questions in the space provided.

9. How do the outer-shell electron configurations for ions of Group 1, Group 2, and Group 15, Group 16, and Group 17 elements compare with those of the noble gases?

They are the same. Group 1 + 2 ions have noble gas configuration of previous row. Group 15, 16 + 17 ions have noble gas configuration in their row.

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JE Gale



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Table of Contents Skills Concept Review Waste Answers

1. Understanding the eBook Skills Concept Review Waste Answers
 - The Rise of Digital Reading Skills Concept Review Waste Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Skills Concept Review Waste Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Skills Concept Review Waste Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Skills Concept Review Waste Answers
 - Personalized Recommendations
 - Skills Concept Review Waste Answers User Reviews and Ratings
 - Skills Concept Review Waste Answers and Bestseller Lists

5. Accessing Skills Concept Review Waste Answers Free and Paid eBooks
 - Skills Concept Review Waste Answers Public Domain eBooks
 - Skills Concept Review Waste Answers eBook Subscription Services
 - Skills Concept Review Waste Answers Budget-Friendly Options
6. Navigating Skills Concept Review Waste Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Skills Concept Review Waste Answers Compatibility with Devices
 - Skills Concept Review Waste Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Skills Concept Review Waste Answers
 - Highlighting and Note-Taking Skills Concept Review Waste Answers
 - Interactive Elements Skills Concept Review Waste Answers
8. Staying Engaged with Skills Concept Review Waste Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Skills Concept Review Waste Answers
9. Balancing eBooks and Physical Books Skills Concept Review Waste Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Skills Concept Review Waste Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Skills Concept Review Waste Answers
 - Setting Reading Goals Skills Concept Review Waste Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Skills Concept Review Waste Answers
 - Fact-Checking eBook Content of Skills Concept Review Waste Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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