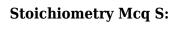
CHAPTER - 1 - MCQS

_	Chemistry is chiefly concerned with the study of changes in matter involving
10	
	a. forces d. electricity
	c. composition d. electricity c. p. which represents the greatest mass is one
t.	c. composition The sample of oxygen at S.T.P. which represents the greatest mass is one b. gram
	a. mole
	c. molecule
i i	c. molecule The mass of a mole of a gas divided by 22.4 liters equals
in Process	a. the density of the gas at STP
	b. the volume of the gas
	a the empirical formula of the gas
	. a - aa - 1043 1laa at 100 1198
	d, the volume of 6.02 x 10° molecules of the gas
$\delta_{\rm p}$	The formula CO2 does not give information about
	a. the elements which make up the compound.
	 the number of atoms of each element in the compound.
	c. the solubility of the compound in water.
	d. the relative mass of the molecule.
i	The molecular formula H2O represents all of the following except
333	a. one molecule of water
	a. One more or water
	36

Stoichiometry Mcq S

DP Hallahan



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