

CHAPTER – 1 – MCQS

- i. Chemistry is chiefly concerned with the study of changes in matter involving
 - a. forces
 - b. heat
 - c. composition
 - d. electricity
- i. The sample of oxygen at S.T.P. which represents the greatest mass is one
 - a. mole
 - b. gram
 - c. molecule
 - d. liter
- i. The mass of a mole of a gas divided by 22.4 liters equals
 - a. the density of the gas at STP
 - b. the volume of the gas
 - c. the empirical formula of the gas
 - d. the volume of 6.02×10^{23} molecules of the gas
- i. The formula CO_2 does not give information about
 - a. the elements which make up the compound.
 - b. the number of atoms of each element in the compound.
 - c. the solubility of the compound in water.
 - d. the relative mass of the molecule.
- i. The molecular formula H_2O represents all of the following except
 - a. one molecule of water

Stoichiometry Mcq S

DP Hallahan



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