

Conversion factors:

Molar mass = #g/1 mole

Avogadro's number = 6.02×10^{23} atoms or molecules/1 mole

Molarity = #moles/L

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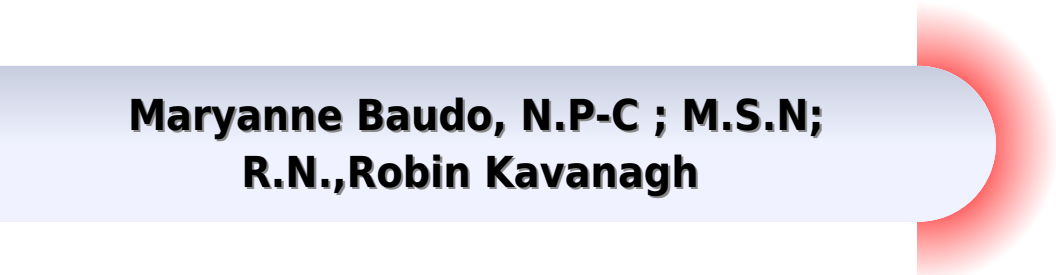
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Stoichiometry Practice Problem Set

1. In the following reaction $4\text{Fe}(s) + 3\text{O}_2(g) \rightarrow 2\text{Fe}_2\text{O}_3(s)$, iron oxidizes to form rust.
 - a) If 40.0g of iron react, how many moles of Fe_2O_3 iron (III) oxide are produced?
 - b) If 20.0g of iron react, how many grams of Fe_2O_3 are produced?
 - c) If 10.0 moles of iron react, how many molecules of Fe_2O_3 are produced?
 - d) If 3.00×10^{22} atoms of iron react, how many moles of Fe_2O_3 are produced?
2. In the reaction for photosynthesis, $6\text{CO}_2(g) + 6\text{H}_2\text{O}(l) + \text{light energy} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6(aq) + 6\text{O}_2(g)$,
 - a) If 715g of CO_2 gas is consumed, how many grams of glucose ($\text{C}_6\text{H}_{12}\text{O}_6$) are produced? The molar mass of glucose is 180.16g/mol.
 - b) How many molecules of carbon dioxide are required to produce 5.9×10^{24} molecules of oxygen gas?

Stoichiometry Guided Practice Problems

**Maryanne Baudo, N.P-C ; M.S.N;
R.N.,Robin Kavanagh**



Stoichiometry Guided Practice Problems:

The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises

A Visual Analogy Guide to Chemistry, 2e Paul A Krieger, 2018-02-01 A Visual Analogy Guide to Chemistry is the latest in the innovative and widely used series of books by Paul Krieger This study guide delivers a big picture view of difficult concepts and effective study tools to help students learn and understand the details of general organic and biochemistry topics A Visual Analogy Guide to Chemistry is a worthwhile investment for any introductory chemistry student

Survival Guide to General Chemistry Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics The focus is not to recap or review the theoretical concepts well described in the available texts Instead the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry Explanations instructional process sequences solved examples and completely solved practice problems are greatly expanded containing significantly more detail than can usually be devoted to in a comprehensive text Many chapters also provide alternative viewpoints as an aid to understanding Key Features The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester Each is written in a specific and detailed step by step process for problem solving whether mathematical or conceptual Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts Includes a chapter designed to eliminate confusion concerning acid base reactions which often persists through working with acid base equilibrium Many chapters provide alternative viewpoints as an aid to understanding This book addresses a very real need for a large number of incoming freshman in STEM fields

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Resources in Education, 1993 *The Extraordinary Chemistry of Ordinary Things, Study Guide* Carl H. Snyder, David F. Dever, 1995-01-19 Examines the chemistry of the substances of our everyday world Our daily lives are immersed in chemicals an effective way to teach and learn chemistry is by examining the goods and substances that we use in our daily lives and that affect us and our environment *The Practice of Chemistry* Donald J. Wink, Sharon Fetzner-Gislason, Sheila McNicholas, 2003-03 Students can't do chemistry if they can't do the math The Practice of Chemistry First Edition is the only preparatory chemistry text to offer students targeted consistent mathematical support to make sure they understand how to use math especially algebra in chemical problem solving The book's unique focus on actual chemical practice extensive study tools and integrated media makes The Practice of Chemistry the most effective way to prepare students for the standard general chemistry course and bright futures as science majors This special PowerPoint tour of the text was created by Don Wink http://www.bfwpub.com/pdfs/wink/POCPowerPoint_Final.ppt 832KB *The Complete Idiot's Guide to Nursing Entrance*

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