

STOICHIOMETRY PRACTICE WORKSHEET

1. Using the following equation:



Calculate how many grams of iodine are needed to prepare 28.6 grams of ICl by this reaction.

2. Using the following equation:



How many moles and grams of KMnO_4 are needed for this reaction on 11.4 grams of KNO_2 ?

3. Using the following equation:



How many moles and grams of oxygen (O_2) are needed to react with 56.8 grams of ammonia by this reaction?

4. Using the following equation:



Calculate the number of moles and the number of grams of iodine (I_2) that can be made this way from 16.4 grams of NaIO_3 .

Stoichiometry Guided Practice Problem 25 Answer

John S. Phillips, Cheryl Wistrom



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The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises

Problems Of Instrumental Analytical Chemistry: A Hands-on Guide (Second Edition) Jose Manuel Andrade Garda, Alatzne Carlosena-zubieta, Marcos Gestal-pose, Maria Paz Gomez-carracedo, Miguel A Maestro-saavedra, Maria Del Carmen Prieto-blanco, Rosa Maria Soto-ferreiro, Javier Teran-baamonde, 2025-07-02 The complex field of analytical chemistry requires knowledge and application of the fundamental principles of numerical calculation Problems of Instrumental Analytical Chemistry provides support and guidance to help students develop these numerical strategies to generate information from experimental results in an efficient and reliable way The book contains exercises that provide standard protocols for the most common calculations in the daily work of a laboratory Also included are easy to follow diagrams to facilitate understanding and avoid common errors making this textbook perfect as a hands on accompaniment to in class learning The subjects covered follow a course in analytical chemistry from the initial basics of data analysis to applications of mass UV VIS infrared and atomic spectrometry and chromatography concluding with an overview of nuclear magnetic resonance and electrochemistry Intended as a self training tool for undergraduates in chemistry analytical chemistry and related subjects this book is also useful as a reference for scientists looking to brush up on their knowledge of instrumental techniques in laboratories This second edition builds upon the first with new and updated content as well as QR codes distributed throughout directing readers to dedicated materials and websites hosting additional information examples and models

Learning to Solve Problems David H. Jonassen, 2004-05-03 Learning to Solve Problems is a much needed book that describes models for designing interactive learning environments to support how to learn and solve different kinds of problems Using a research based approach author David H Jonassen a recognized expert in the field shows how to design instruction to support three kinds of problems story problems troubleshooting and case and policy analysis problems Filled with models and job aids this book describes different approaches for representing problems to learners and includes information about technology based tools that can help learners mentally represent problems for themselves Jonassen also explores methods for associating different solutions to problems and discusses various processes for reflecting on the problem solving process Learning to Solve Problems also includes three methods for assessing problem solving skills performance assessment component skills and argumentation

Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998

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