

Stoichiometry Review

1. The combustion of propane occurs via the reaction:



How many grams of oxygen are required to burn completely 10.0 g of propane?

2. Small quantities of chlorine can be prepared in the laboratory by the reaction:



What volume of chlorine at STP can be prepared from 100.0 g of manganese dioxide?

3. The most common ore of arsenic is mispickel, FeSAs . Upon heating this ore, free arsenic is obtained. How many grams of FeSAs are required to produce 10.0 g of arsenic according to the following reaction?



4. Glucose is used as an energy source by the human body. The overall reaction in the body is:



Calculate the volume of oxygen at STP required to convert 28.0 g of glucose to carbon dioxide and water.

5. Potassium nitrate is widely used as a fertilizer because it provides two essential elements, potassium and nitrogen. It is made by mixing potassium chloride and nitric acid in the presence of oxygen according to the equation:



How many kilograms of potassium nitrate will be produced from 50.0 kg of potassium chloride and 50.0 kg of nitric acid with excess oxygen?

6. Bromine can be prepared by adding chlorine to an aqueous solution of sodium bromide. How many grams of bromine are formed if 25.0 g of chlorine and 25.0 g of sodium bromide are reacted according to the reaction below? Which reactant is in excess and how much remains after the reaction is complete?



7. How many millilitres of 2.00 M hydrochloric acid are required to react with 2.55 g of zinc according to the following reaction:



8. A 0.473 g sample of phosphorus is reacted with an excess of chlorine. Calculate the mass of phosphorus pentachloride produced.

9. How many grams of oxygen can be prepared by the decomposition of 25.0 g of mercury(II) oxide?

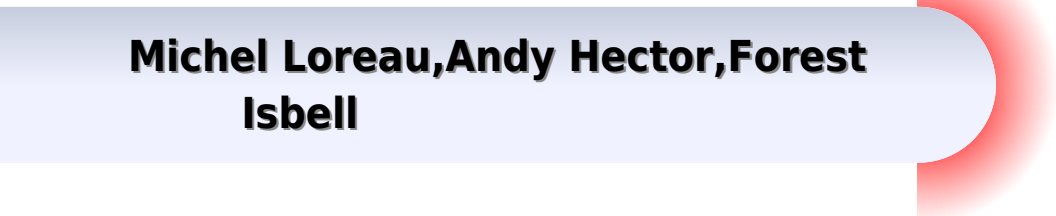
10. What volume of oxygen gas at STP can be prepared by the decomposition of 25.0 g of potassium chlorate?

11. How many grams of zinc are required for the replacement of 0.100 g of hydrogen from sulfuric acid?

12. What mass of magnesium chloride is required to react with 10.0 g of silver nitrate?

Stoichiometry Review Ws

**Michel Loreau, Andy Hector, Forest
Isbell**



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directed at researchers and professionals in plant molecular biology biochemistry and biotechnology The Enzymes of Biological Membranes Anthony Martonosi, 2013-03-09 In the first edition of The Enzymes of Biological Membranes published in four volumes in 1976 we collected the mass of widely scattered information on membrane linked enzymes and metabolic processes up to about 1975 This was a period of transition from the romantic phase of membrane biochemistry preoccupied with conceptual developments and the general properties of membranes to an era of mounting interest in the specific properties of membrane linked enzymes analyzed from the viewpoints of modern enzymology The level of sophistication in various areas of membrane research varied widely the structures of cytochrome c and cytochrome b were known to atomic detail while the majority of membrane linked enzymes had not even been isolated In the intervening eight years our knowledge of membrane linked enzymes expanded beyond the wildest expectations The purpose of the second edition of The Enzymes of Biological Membranes is to record these developments The first volume describes the physical and chemical techniques used in the analysis of the structure and dynamics of biological membranes In the second volume the enzymes and metabolic systems that participate in the biosynthesis of cell and membrane components are discussed The third and fourth volumes review recent developments in active transport oxidative phosphorylation and photosynthesis *EJB Reviews* 1996 P. Christen, E. Hofmann, 2012-12-06 In the mid 1980s the European Journal of Biochemistry set out to publish review articles The enterprise proved successful resulting in high level reviews written by well known scientists appearing in the Journal The reviews represent emerging and rapidly growing fields of research in fundamental as well as applied areas of biochemistry such as medicine biotechnology agriculture and nutrition Novel methodological and technological approaches which stimulate biochemical research are also included The authors of the reviews are explicitly asked to be critical selective evaluative and interdisciplinary oriented The reviews should encourage young scientists to think independently and creatively and inform active investigators about the state of the art in a given field **Advances in Catalysis** , 1998-05-08 Since 1948 this Series has filled the gap between the papers that report and the textbooks that teach in the diverse areas of catalysis research The editors of and contributors to Advances in Catalysis are dedicated to recording progress in this area Each volume of Advances in Catalysis contains articles covering a subject of broad interest *Annual Review of Plant Physiology and Plant Molecular Biology* Russell L. Jones, 1999-06 *Advances in Synthesis Gas: Methods, Technologies and Applications* Mohammad Reza Rahimpour, Mohammad Amin Makarem, Maryam Meshksar, 2022-10-18 Advances in Synthesis Gas Methods Technologies and Applications Syngas Production and Preparation is a collection of various chapters concerning many aspects of syngas production technologies including common methods like gasification steam dry autothermal reforming membrane technology etc along with novel methods like plasma technology micro reactors electrolysis processes as well as photocatalytic systems In addition different sources for producing syngas including oil crude oil heavy oil microalgae black liquor tar and bitumen as well as municipal agricultural food plastic wood and cardboard

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