

The Advanced Placement Examination in Chemistry

Part I – Multiple Choice Questions

Part II - Free Response Questions

Selected Questions from 1970 to 2010

Stoichiometry

Part I

1984

2. Which of the following forms a compound having the formula KNO_3 ?
- (A) F
(B) S
(C) Mg
(D) Ar
(E) Mn
32. The net ionic equation for the reaction between silver carbonate and hydrochloric acid is
- (A) $\text{Ag}_2\text{CO}_3(s) + 2 \text{H}^+ + 2 \text{Cl}^- \rightarrow 2 \text{AgCl}(s) + \text{H}_2\text{O} + \text{CO}_2(g)$
(B) $2 \text{Ag}^+ + \text{CO}_3^{2-} + 2 \text{H}^+ + 2 \text{Cl}^- \rightarrow 2 \text{AgCl}(s) + \text{H}_2\text{O} + \text{CO}_2(g)$
(C) $\text{CO}_3^{2-} + 2 \text{H}^+ \rightarrow \text{H}_2\text{O} + \text{CO}_2(g)$
(D) $\text{Ag}^+ + \text{Cl}^- \rightarrow \text{AgCl}(s)$
(E) $\text{Ag}_2\text{CO}_3(s) + 2 \text{H}^+ \rightarrow 2 \text{Ag}^+ + \text{H}_2\text{CO}_3$
44. What number of moles of O_2 is needed to produce 14.2 grams of P_2O_{10} from P? (Molecular weight $\text{P}_2\text{O}_{10} = 284$)
- (A) 0.0500 mole
(B) 0.0625 mole
(C) 0.125 mole
(D) 0.250 mole
(E) 0.500 mole
45. The alkenes are compounds of carbon and hydrogen with the general formula C_nH_{2n} . If 0.561 gram of any alkene is burned in excess oxygen, what number of moles of H_2O is formed?
- (A) 0.0400 mole
(B) 0.0600 mole
(C) 0.0800 mole
(D) 0.400 mole
(E) 0.800 mole
52. $3 \text{Ag}(s) + 4 \text{HNO}_3 \rightarrow 3 \text{AgNO}_3 + \text{NO}(g) + 2 \text{H}_2\text{O}$
The reaction of silver metal and dilute nitric acid proceeds according to the equation above. If 0.10 mole of powdered silver is added to 10. milliliters of 6.0-molar nitric acid, the number of moles of NO gas that can be formed is
- (A) 0.015 mole
(B) 0.020 mole
(C) 0.030 mole
(D) 0.045 mole
(E) 0.090 mole

Stoichiometry Multiple Choice Questions

Jiyuan Zhang



Stoichiometry Multiple Choice Questions:

The book delves into Stoichiometry Multiple Choice Questions. Stoichiometry Multiple Choice Questions is a crucial topic that must be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Stoichiometry Multiple Choice Questions, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Stoichiometry Multiple Choice Questions
- Chapter 2: Essential Elements of Stoichiometry Multiple Choice Questions
- Chapter 3: Stoichiometry Multiple Choice Questions in Everyday Life
- Chapter 4: Stoichiometry Multiple Choice Questions in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Stoichiometry Multiple Choice Questions. This chapter will explore what Stoichiometry Multiple Choice Questions is, why Stoichiometry Multiple Choice Questions is vital, and how to effectively learn about Stoichiometry Multiple Choice Questions.
3. In chapter 2, this book will delve into the foundational concepts of Stoichiometry Multiple Choice Questions. This chapter will elucidate the essential principles that must be understood to grasp Stoichiometry Multiple Choice Questions in its entirety.
4. In chapter 3, the author will examine the practical applications of Stoichiometry Multiple Choice Questions in daily life. This chapter will showcase real-world examples of how Stoichiometry Multiple Choice Questions can be effectively utilized in everyday scenarios.
5. In chapter 4, this book will scrutinize the relevance of Stoichiometry Multiple Choice Questions in specific contexts. The fourth chapter will explore how Stoichiometry Multiple Choice Questions is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, the author will draw a conclusion about Stoichiometry Multiple Choice Questions. This chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Stoichiometry Multiple Choice Questions.

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