



Name: Chelsey Williams Date: 3/31/22

Student Exploration: Stoichiometry

Vocabulary: Avogadro's number, balanced equation, cancel, coefficient, conversion factor, dimensional analysis, formula mass, molar mass, mole, molecular mass, stoichiometry

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. A 250 mL glass of orange juice contains 22 grams of sugar. How much sugar is in a two-liter (2,000 mL) bottle of orange juice? 176 grams
2. It requires two sticks of butter to make a batch of 20 cookies. How much butter will it take to make 150 cookies? 15 sticks of butter

Gizmo Warm-up

Just as a cook follows a recipe to decide how much of each ingredient to add, a chemist uses **stoichiometry** to determine the amounts of substances involved in chemical reactions. The Stoichiometry Gizmo allows you to try your hand at figuring out the amounts of reactants and products that take part in a chemical reaction.

To begin, check that this equation is shown:



1. Look at the **coefficients** (such as the "3" in 3CO) in front of each substance in the equation. The coefficients tell you how many molecules or atoms take part in a chemical reaction. In the spaces below, list the number of each molecule or atom in the equation:

Fe_2O_3 1

CO 3

Fe 2

CO_2 3

2. In a **balanced equation**, the same number of each kind of atom is shown on each side of the equation. Calculate the number of iron (Fe), oxygen (O), and carbon atoms (C).

New equation



New question

How many moles of carbon monoxide (CO) are required to react completely with 1.75 moles of iron (III) oxide (Fe_2O_3)?

Problems solved: 0



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K Morrison



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Food Harald Brüssow, 2007-05-11 This book explores the links between food and human cultural and physical evolution Each chapter begins by summarizing the basic knowledge in the field discusses recent research results and confirms or challenges established concepts inviting new insight and provoking new questions This book catalyzes discussion between scientists

working on one side in food science and on the other side in biological and biomedical research **Choice**, 2002 [Metric Myth & Quasicrystals](#) Antony J. Bourdillon, 2012-08 Metric Myth the series is not properly Fibonacci simplified indexation of the pattern is used to simulate a structure due to a single aligned edge sharing unit cell that is consistent with all data Since it is unlikely that the sharp diffraction pattern is due to unmeasured poly polyhedra does the International Union of Crystallography have to redefine crystals yet again In modern physics the metric relates the covariant components of invariant vectors with corresponding contravariant components In crystallography it relates dimensions in momentum space to atomic locations in real space In quasicrystals the pattern in momentum space is logarithmic Theory and simulation show why this has to be Consequences follow In particular we show not only where the atoms are but also why they are there A debate is reported so that the reader will be encouraged to make his own mind When logarithmic periodicity is discovered and explained in one branch of physics it should be expected in others

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The Publishers' Trade List Annual, 1974

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