

Chapter 10: Stoichiometry

Stoichiometry

Teacher Notes and Answers

Question 1:
Moles of H_2SO_4

Answer: 0.0025 moles

- 10.1
- 10.2
- 10.3
- 10.4

10.5: a. 0.0025 moles, b. 0.0025 moles

10.6: a. 0.0025 moles, b. 0.0025 moles

10.7: a. 0.0025 moles, b. 0.0025 moles

10.8: a. 0.0025 moles, b. 0.0025 moles

10.9: a. 0.0025 moles, b. 0.0025 moles

10.10: a. 0.0025 moles, b. 0.0025 moles

10.11: a. 0.0025 moles, b. 0.0025 moles

10.12: a. 0.0025 moles, b. 0.0025 moles

10.13: a. 0.0025 moles, b. 0.0025 moles

10.14: a. 0.0025 moles, b. 0.0025 moles

10.15: a. 0.0025 moles, b. 0.0025 moles

10.16: a. 0.0025 moles, b. 0.0025 moles

10.17: a. 0.0025 moles, b. 0.0025 moles

10.18: a. 0.0025 moles, b. 0.0025 moles

10.19: a. 0.0025 moles, b. 0.0025 moles

10.20: a. 0.0025 moles, b. 0.0025 moles

10.21: a. 0.0025 moles, b. 0.0025 moles

10.22: a. 0.0025 moles, b. 0.0025 moles

Question 2:
Moles of H_2SO_4

Answer: 0.0025 moles

- 10.1
- 10.2
- 10.3
- 10.4
- 10.5
- 10.6
- 10.7
- 10.8
- 10.9
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- 10.93
- 10.94
- 10.95
- 10.96
- 10.97
- 10.98
- 10.99
- 10.100

Question 3:
Moles of H_2SO_4

Answer: 0.0025 moles

- 10.1
- 10.2
- 10.3
- 10.4
- 10.5
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- 10.9
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