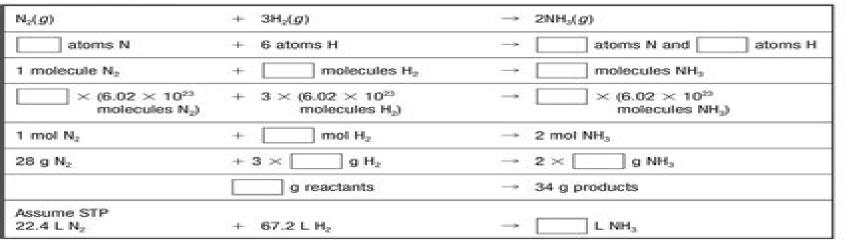
CHAPTER 12, Stoichiometry (continued)

- 8. The coefficients of a balanced chemical equation tell you the relative number of moles of ______ in a chemical reaction.
- 9. Why is the relative number of moles of reactants and products the most important information that a balanced chemical equation provides?

► Mass Conservation in Chemical Reactions (pages 357–358)

- 10. Is the following sentence true or false? A balanced chemical equation must obey the law of conservation of mass.
- Use Figure 12.3 on page 357. Complete the table about the reaction of nitrogen and hydrogen.



- Circle the letter(s) of the items that are ALWAYS conserved in every chemical reaction.
 - a. volume of gases

d. moles

b. mass

e. molecules

c. formula units

- f. atoms
- 13. What reactant combines with oxygen to form sulfur dioxide? Where can this reactant be found in nature?

Stoichiometry Section 1

George Frederick Barker

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