

Quiz: Solution Stoichiometry and Net Ionic Eqn. SCH3U

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Assessment of Learning

1. Write the balanced chemical equation, total ionic equation and net ionic equation for tin (IV) bromide reacting with iron (III) phosphate (60)

2. 250 mL of 0.650 mol/L $\text{Pb}(\text{NO}_3)_2$ is mixed with 345 mL of 0.469 mol/L NaI .
 $\text{Pb}(\text{NO}_3)_2 + \text{NaI} \rightarrow \text{NaNO}_3 + \text{PbI}_2$ (not balanced)
a) Calculate the limiting reactant and the concentration of the sodium nitrate solution (in mol/L) that will be produced. (4T,1C)
b) The amount of excess reactant in grams. (2T,1C)

Stoichiometry Modern Chemistry Section Quiz Answers

Didier Musso



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