

Name: _____

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S'mores Stoichiometry Lab

Introduction:

In this experiment, you will explore the principles of stoichiometry by building S'mores, the delicious, chocolate, marshmallow, and graham cracker treats.



Definitions: 'Stoichio' means element and 'metry' means the process of measuring. The mass and quantity relationships among reactants and products in a reaction are found using the process of stoichiometry.

Purpose:

To determine the limiting reactant in the synthesis of S'mores.

Prelab Questions:

1. What does a chemical equation tell us?
2. What do the coefficients in a balanced chemical equation represent?

Materials:

Miniature Chocolate bars
Marshmallows
Bunsen Burner

Graham cracker Square
Paper Towels
Wooden skewers

Reference Data:

Substance	Symbol	Unit mass	Package Mass
Graham Cracker Squares	Gr	8.50 g	408 g
Marshmallow	Mm	7.125 g	285 g
Miniature Chocolate Bar	Ch	12.00 g	348 g

** for the purposes of this lab the UNIT mass and MOLAR mass are the same thing

Procedure:

1. Perform a synthesis reaction (forming ~~one~~ s'more). For this lab, we will consider a s'more to consist of 2 graham cracker squares, 1 marshmallow and 1 miniature chocolate bar.
 - a) Determine the formula for a s'more
 - b) write the balanced chemical equation using the symbols above
 - c) According to the balanced equation complete the following ratios:

_____ Gr = _____ Mm	_____ Gr = _____ S'more
_____ Ch = _____ Mm	_____ Mm = _____ S'more
_____ Ch = _____ Gr	_____ Ch = _____ S'more
2. Calculating the unit mass of the s'more.
 - a) Calculate the mass on ONE s'more.
 - b) Is there a relationship between the mass of a S'more and the masses of the reactants used to make it?
 - c) If so, what is the relationship?
 - d) What law have you studied in this course that might define this relationship?
3. Predicting the limiting reactant:
 - a) Complete the data table below based on the materials provided for your group.

Substance	Quantity received	# of S'mores it COULD make
Graham Cracker Squares		
Marshmallow		
Miniature Chocolate Bar		
 - b) What will the limiting reactant be?
 - c) How many of the other substances will remain?
 - d) What are the masses of each excess reactant that will remain once the reaction is complete?

Stoichiometry Lab Answer Key Aluminum

Yi-Tong Ma

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Stoichiometry Lab Answer Key Aluminum:

This book delves into Stoichiometry Lab Answer Key Aluminum. Stoichiometry Lab Answer Key Aluminum is an essential topic that needs to be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Stoichiometry Lab Answer Key Aluminum, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Stoichiometry Lab Answer Key Aluminum
- Chapter 2: Essential Elements of Stoichiometry Lab Answer Key Aluminum
- Chapter 3: Stoichiometry Lab Answer Key Aluminum in Everyday Life
- Chapter 4: Stoichiometry Lab Answer Key Aluminum in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Stoichiometry Lab Answer Key Aluminum. This chapter will explore what Stoichiometry Lab Answer Key Aluminum is, why Stoichiometry Lab Answer Key Aluminum is vital, and how to effectively learn about Stoichiometry Lab Answer Key Aluminum.
3. In chapter 2, the author will delve into the foundational concepts of Stoichiometry Lab Answer Key Aluminum. The second chapter will elucidate the essential principles that must be understood to grasp Stoichiometry Lab Answer Key Aluminum in its entirety.
4. In chapter 3, the author will examine the practical applications of Stoichiometry Lab Answer Key Aluminum in daily life. This chapter will showcase real-world examples of how Stoichiometry Lab Answer Key Aluminum can be effectively utilized in everyday scenarios.
5. In chapter 4, the author will scrutinize the relevance of Stoichiometry Lab Answer Key Aluminum in specific contexts. The fourth chapter will explore how Stoichiometry Lab Answer Key Aluminum is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, the author will draw a conclusion about Stoichiometry Lab Answer Key Aluminum. This chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Stoichiometry Lab Answer Key Aluminum.

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