

## Unit 6: Stoichiometry Homework Packet

Name: \_\_\_\_\_

Period: \_\_\_\_\_

### Mole Ratios:

1. What is a mole ratio?

\_\_\_\_\_

2. The coefficients in a chemical equation indicate the relationship among \_\_\_\_\_ of reactants and products.

3. Write 3 possible mole ratios from this equation:  $4\text{Fe} + 3\text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3$

4. Write 4 possible mole ratios from this equation:  $\text{C}_2\text{H}_8 + 5\text{O}_2 \rightarrow 3\text{CO}_2 + 4\text{H}_2\text{O}$

5. Write a balanced chemical equation when nitrogen reacts with hydrogen to produce nitrogen trihydride. Determine 3 possible mole ratios.

6. Stoichiometry is based on the \_\_\_\_\_.

7. This law states that \_\_\_\_\_.

**Write the mole ratio in the space provided and solve each problem.**



8. How many moles of sodium hydroxide are produced when 1.50 mol sodium peroxide reacts with water?

$$1.50 \text{ mol Na}_2\text{O}_2 \quad \times \quad \frac{4 \text{ mol NaOH}}{2 \text{ mol Na}_2\text{O}_2} \quad = \quad \underline{3.00} \quad \text{mol NaOH}$$

9. How many moles of water are required to produce 2.15 mol oxygen gas in this reaction?

$$2.15 \text{ mol O}_2 \quad \times \quad \frac{2 \text{ mol H}_2\text{O}}{1 \text{ mol O}_2} \quad = \quad \underline{4.30} \quad \text{mol H}_2\text{O}$$

**Stoichiometry Practice Problems #1..... Solve each problem, show all work.**

1. How many **grams** of  $\text{Cl}_2$  must be reacted with excess  $\text{NaI}$  if 10.0g of  $\text{NaCl}$  are needed?



# Stoichiometry Homework Packet On Moles

**Ardas Ozsogomonyan**



## **Stoichiometry Homework Packet On Moles:**

*The Practice of Chemistry Study Guide & Solutions Manual* Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises

**Chemistry** James N. Spencer, George M. Bodner, Lyman H. Rickard, 2010-12-28 CHEMISTRY The Big Book of Chemistry Teacher Stories Jeff Lark, Stories from years of teaching high school chemistry *Chapter-wise DPP Sheets for Chemistry JEE Advanced* Disha Experts, 2019-10-10 The book Chapter wise Daily Practice Problem DPP Sheets for Chemistry JEE Advanced contains 1 Carefully selected Questions of all new pattern variety 20 per DPP in Chapter wise DPP Sheets for Practice At the end one Full Test is provided 2 The book is divided into 26 Chapter wise DPPs based on the NCERT 3 Time Limit Maximum Marks Cutoff Qualifying Score for each DPP Sheet is provided 4 These sheets will act as an Ultimate tool for Concept Checking Speed Building 5 Collection of 560 MCQ s of all variety of new pattern 6 Covers all important Concepts of each Chapter 7 As per latest pattern syllabus of JEE Advanced exam

**Solving Problems in Chemistry** Rod O'Connor, Charles Mickey, Alton Hassell, 1977

**Three Cognitive Skills in Chemistry and Their Application to Stoichiometry** Ardas Ozsogomonyan, 1977

Chemistry Greg Curran, 2011 Covers all the topics in a typical one year high school chemistry curriculum (Free Sample) GO TO Objective NEET Chemistry Guide with DPP & CPP Sheets 9th Edition Disha Experts, 2021-10-07 The thoroughly revised updated 9th Edition of Go To Objective NEET Chemistry is developed on the objective pattern following the chapter plan as per the NCERT books of class 11 and 12 The book has been rebranded as GO TO keeping the spirit with which this edition has been designed The complete book has contains 31 Chapters In the new structure the book is completely revamped with every chapter divided into 2 4 Topics Each Topic contains Study Notes along with a DPP Daily Practice Problem of 15 20 MCQs This is followed by a Revision Concept Map at the end of each chapter The theory is followed by a set of 2 Exercises for practice The first exercise is based on Concepts Application It also covers NCERT based questions This is followed by Exemplar past 8 year NEET 2013 2021 questions In the end of the chapter a CPP Chapter Practice Problem Sheet of 45 Quality MCQs is provided The solutions to all the questions have been provided immediately at the end of each chapter

Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998

*The Latest and Best of TESS*, 1991

A Stoichiometry Unit David Callaghan, 2004

**Introduction to Chemical Principles** Howard Stephen Stoker, 1990 Introduction to Chemical Principles is a text for students who have had little to no previous instruction in chemistry or who had such instruction long enough ago that a thorough review is needed preface

*Resources in Education*, 1997-05

**Chemistry: Po-Z** J. J. Lagowski, 2004 This is a reference tool designed to guide the reader through all the aspects of chemistry Showing the myriad of ways in which chemistry plays a role both seen and unseen in our daily lives this work also makes the foundations of chemistry accessible for the lay reader

*The Software Finder*, 1983

**The Practice of Chemistry** Donald J. Wink, Sharon Fetzer-Gislason, Sheila McNicholas, 2003-03 Students

can't do chemistry if they can't do the math The Practice of Chemistry First Edition is the only preparatory chemistry text to offer students targeted consistent mathematical support to make sure they understand how to use math especially algebra in chemical problem solving The book's unique focus on actual chemical practice extensive study tools and integrated media makes The Practice of Chemistry the most effective way to prepare students for the standard general chemistry course and bright futures as science majors This special PowerPoint tour of the text was created by Don Wink [http://www.bfwpub.com/pdfs/wink/POCPowerPoint\\_Final.ppt](http://www.bfwpub.com/pdfs/wink/POCPowerPoint_Final.ppt) 832KB

**Chemistry** Neil D. Jespersen, Alison Hyslop, 2021-11-02 Chemistry The Molecular Nature of Matter 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter Key revisions in this edition focus on three areas The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry Simultaneously examples and questions have been updated to align them with career concepts relevant to the environmental engineering biological pharmaceutical and medical sciences Providing students with transferable skills with a focus on integrating metacognition and three dimensional learning into the text When students know what they know they are better able to learn and incorporate the material Providing a total solution through New WileyPLUS by fully integrating the enhanced eText with online assessment answer specific responses and additional practice resources The 8th edition continues to emphasize the importance of applying concepts to problem solving to achieve high level learning and increase retention of chemistry knowledge Problems are arranged in an intuitive confidence building order

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Reactive Transport Modeling Yitian Xiao, Fiona Whitaker, Tianfu Xu, 2018-06-05 Teaches the application of Reactive Transport Modeling RTM for subsurface systems in order to expedite the understanding of the behavior of complex geological systems This book lays out the basic principles and approaches of Reactive Transport Modeling RTM for surface and subsurface environments presenting specific workflows and applications The techniques discussed are being increasingly commonly used in a wide range of research fields and the information provided covers fundamental theory practical issues in running reactive transport models and how to apply techniques in specific areas The need for RTM in engineered facilities such as nuclear waste repositories or CO<sub>2</sub> storage sites is ever increasing because the prediction of the future evolution of these systems has become a legal obligation With increasing recognition of the power of these approaches and their widening adoption comes responsibility to ensure appropriate application of available tools This book aims to provide the requisite understanding of key aspects of RTM and in doing so help identify and thus avoid potential pitfalls Reactive Transport Modeling covers the application of RTM for CO<sub>2</sub> sequestration and geothermal energy development reservoir quality prediction modeling diagenesis modeling geochemical processes in oil modeling gas hydrate production reactive transport in fractured and porous media reactive transport studies for nuclear waste disposal reactive flow modeling in hydrothermal systems and modeling biogeochemical processes Key features include A comprehensive reference for scientists and practitioners entering the area of reactive transport modeling RTM Presented by internationally known experts in the field Covers fundamental theory practical issues in running reactive transport models and hands on examples for applying techniques in specific areas Teaches readers to appreciate the power of RTM and to stimulate usage and application Reactive Transport Modeling is written for graduate students and researchers in academia government laboratories and industry who are interested in applying reactive transport modeling to the topic of their research The book will also appeal to geochemists hydrogeologists geophysicists earth scientists environmental engineers and environmental chemists

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