

## Chapter 9 Study Guide

### Holt Chemistry: Stoichiometry

#### Questions

1. What is stoichiometry? Finding mass of product formed (or reactant consumed) when given a mass of another species in the reaction, by using the mole to mole ratio from the balanced equation.
2. A balanced equation verifies the law of conservation of mass: Total mass of reactants must equal total mass of products.
3. Give the 3 steps of converting the mass (grams) of one substance to the mass (grams) of another substance: 1. Convert grams to moles using molar mass; 2. Mole to Mole ratio from balanced equation; 3. Convert moles back to grams using molar mass

# Stoichiometry 9 1 Answers Holt Mcdougal

**Holt Rinehart & Winston, Holt,  
Rinehart and Winston Staff**



**Stoichiometry 9 1 Answers Holt Mcdougal:**

Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001

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