

Gas Stoichiometry: Multiple Choice

1. What volume of pure oxygen gas, O_2 , measured at 546 K and 101.325 kPa is formed by the complete decomposition of 0.200 mol Ag_2O according to the following equation:



- A) 1.12 L B) 4.48 L C) 2.24 L D) 8.96 L E) 17.82 L

ANS: B

1. How many moles of $KClO_3$ are needed to form 2.8 L of O_2 , measured at STP, according to the following reaction: $2KClO_3 \rightarrow 2KCl + 3O_2$

- A) 0.083 mol B) 0.25 mol C) 0.17 mol D) 0.50 mol E) 0.33 mol

ANS: A

1. What volume of chlorine gas, measured at 0°C and 101.325 kPa, is required to react completely with 5.61 g of KOH in the following reaction: $Cl_{2(g)} + 2KOH \rightarrow KCl + KClO + H_{2O}$

- A) 0.56 L B) 4.48 L C) 1.12 L D) 22.4 L E) 2.24 L

ANS: C

1. Sodium hydride reacts with excess water to produce sodium hydroxide (aq) and hydrogen gas: $NaH(s) + H_2O(l) \rightarrow NaOH(aq) + H_2(g)$

A sample of NaH weighing _____ g will produce 982 mL of gas at 28.0 °C and 765 torr, when the hydrogen is collected over water. The vapor pressure of water at this temperature is 28 torr.

- A) 2.93 B) 0.960 C) 0.925 D) 0.0388 E) 925

ANS: C

1. At STP, how many liters of oxygen are required to react completely with 3.6 liters of hydrogen to form water? $2H_2(g) + O_2(g) \rightarrow 2H_2O(g)$

- A) 1.8 L B) 3.6 L C) 2.0 L D) 2.4 L

ANS: A

1. The molar volume of a gas at STP is _____ L.

- A) 0.08206 B) 62.36 C) 1.00 D) 22.4 E) 14.7

ANS: D

1. If 0.5 L of $O_2(g)$ reacts with H_2 to produce 1 L of $H_2O(g)$, what is the volume of $H_2O(g)$ obtained from 1 L of $O_{2(g)}$?

- A) 0.5 L B) 1.5 L C) 2 L D) 2.5 L

ANS: C

Stoichiometry Test Multiple Choice Modern Chemistry

Yi-Tong Ma



Stoichiometry Test Multiple Choice Modern Chemistry:

This book delves into Stoichiometry Test Multiple Choice Modern Chemistry. Stoichiometry Test Multiple Choice Modern Chemistry is an essential topic that needs to be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Stoichiometry Test Multiple Choice Modern Chemistry, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Stoichiometry Test Multiple Choice Modern Chemistry
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 - Chapter 4: Stoichiometry Test Multiple Choice Modern Chemistry in Specific Contexts
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- The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Stoichiometry Test Multiple Choice Modern Chemistry.

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