

Stoichiometry



Name: _____

Section: _____



1. Consider the following balanced reaction equation:



- a. What are the following molar ratios:
- mol Na_2CO_3 /mol H_3PO_3 _____
 - mol Na_2CO_3 /mol H_2CO_3 _____
 - mol Na_3PO_3 /mol H_2CO_3 _____
 - mol H_3PO_3 /mol Na_3PO_3 _____
- b. Calculate the number of moles of carbonic acid that can be made from 0.833 moles phosphorous acid.
- c. How many moles of sodium carbonate are needed to react with 3.55×10^{-3} mol of phosphorous acid?
- d. How many moles sodium phosphite can be produced with 0.075 moles of carbonic acid?

Stoichiometry Section Quiz

CO Houle



Stoichiometry Section Quiz:

The Enigmatic Realm of **Stoichiometry Section Quiz**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing short of extraordinary. Within the captivating pages of **Stoichiometry Section Quiz** a literary masterpiece penned with a renowned author, readers set about a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting effect on the hearts and minds of people who partake in its reading experience.

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