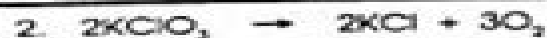


**STOICHIOMETRY:
MOLE-MOLE PROBLEMS**

How many moles of hydrogen are needed to completely react with two moles of nitrogen?

$$2 \text{ mol N}_2 \times \frac{3 \text{ mol H}_2}{1 \text{ mol N}_2} = 6$$

6 moles H₂

How many moles of oxygen are produced by the decomposition of six moles of potassium chlorate?

$$6 \text{ moles KClO}_3 \times \frac{3 \text{ mol O}_2}{2 \text{ mol KClO}_3} = \frac{18}{2} = 9$$

9 moles of O₂

How many moles of hydrogen are produced from the reaction of three moles of zinc with an excess of hydrochloric acid?

$$3 \text{ mol Zn} \times \frac{1 \text{ mol H}_2}{1 \text{ mol Zn}} = 3$$

3 moles of H₂How many moles of oxygen are necessary to react completely with four moles of propane (C₃H₈)?

$$4 \text{ mol C}_3\text{H}_8 \times \frac{5 \text{ mol O}_2}{1 \text{ mol C}_3\text{H}_8} = 20$$

20 mol of O₂

How many moles of potassium nitrate are produced when two moles of potassium phosphate react with two moles of aluminum nitrate?

$$2 \text{ mol K}_3\text{PO}_4 \times \frac{3 \text{ mol KNO}_3}{1 \text{ mol K}_3\text{PO}_4} = 6$$

6 moles of KNO₃

Stoichiometry Introduction Packet Answers

JS Bruner



Stoichiometry Introduction Packet Answers:

This book delves into Stoichiometry Introduction Packet Answers. Stoichiometry Introduction Packet Answers is a crucial topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Stoichiometry Introduction Packet Answers, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Stoichiometry Introduction Packet Answers
 - Chapter 2: Essential Elements of Stoichiometry Introduction Packet Answers
 - Chapter 3: Stoichiometry Introduction Packet Answers in Everyday Life
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