

Unit 08 – Stoichiometry – Worksheet 1

Answer the following questions in the space provided at the left.

- 18 mol O₂ 1. How many moles of oxygen are produced from twelve moles of Potassium chlorate?



$$\frac{12 \text{ mol } \cancel{\text{KClO}_3}}{\cancel{2 \text{ mol } \text{KClO}_3}} \quad | \quad \frac{3 \text{ mol O}_2}{2 \text{ mol } \cancel{\text{KClO}_3}}$$

- 21 mol O₂ 2. How many moles of oxygen are produced from 14 moles of Potassium chlorate?



$$\frac{14 \text{ mol } \cancel{\text{KClO}_3}}{\cancel{2 \text{ mol } \text{KClO}_3}} \quad | \quad \frac{3 \text{ mol O}_2}{2 \text{ mol } \cancel{\text{KClO}_3}}$$

- 45 mol H₂O 3. How many moles of water are produced from 22.5 moles of oxygen?



$$\frac{22.5 \text{ mol } \cancel{\text{O}_2}}{\cancel{1 \text{ mol } \text{O}_2}} \quad | \quad \frac{2 \text{ mol H}_2\text{O}}{1 \text{ mol } \cancel{\text{O}_2}}$$

- 20 mol HCl 4. Calculate the number of moles of Hydrogen chloride produced from 10 moles of hydrogen.



$$\frac{10 \text{ mol } \cancel{\text{H}_2}}{\cancel{1 \text{ mol } \text{H}_2}} \quad | \quad \frac{2 \text{ mol HCl}}{1 \text{ mol } \cancel{\text{H}_2}}$$

- 21 mol Cl₂ 5. Calculate the number of moles of chlorine needed to form 14 moles of Iron (III) chloride.



$$\frac{14 \text{ mol } \cancel{\text{FeCl}_3}}{\cancel{2 \text{ mol } \text{FeCl}_3}} \quad | \quad \frac{3 \text{ mol Cl}_2}{2 \text{ mol } \cancel{\text{FeCl}_3}}$$

Stoichiometry Guided And Study Workbook Answers

Karin Nielsen-Saines



Stoichiometry Guided And Study Workbook Answers:

Adopting the Tune of Expression: An Psychological Symphony within **Stoichiometry Guided And Study Workbook Answers**

In a world consumed by screens and the ceaseless chatter of quick interaction, the melodic elegance and mental symphony developed by the prepared term often fade into the background, eclipsed by the persistent noise and interruptions that permeate our lives. But, situated within the pages of **Stoichiometry Guided And Study Workbook Answers** a charming fictional value filled with fresh emotions, lies an immersive symphony waiting to be embraced. Crafted by an elegant composer of language, that fascinating masterpiece conducts viewers on a psychological trip, skillfully unraveling the concealed tunes and profound affect resonating within each carefully crafted phrase. Within the depths of the touching analysis, we will explore the book is key harmonies, analyze its enthralling writing style, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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