

Stock System



Rules for using the stock system to name Covalent Compounds:

- 1) Place and Name the least electronegative element first
ex: SO_2 not O_2S
- 2) Determine the Oxidation state of the 1st named element
ex: SO_2 $\text{S} = +4$
- 3) Use Roman Numerals to designate the oxidation number
ex: SO_2 is Sulfur (IV)
- 4) Ending for the More electronegative element is -ide
ex: SO_2 is Sulfur (IV) Oxide

Stock System Chemistry

CL Gary



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