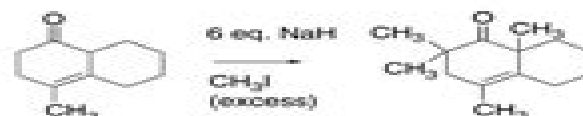


- b. This transformation involves methylation at all enolizable positions. The alkylation was effected using a sixfold excess of NaH and excess methyl iodide. Evidently there is not a significant amount of methylation at C(4), which could occur through γ -alkylation of the C(8a)-enolate.



- c. This alkylation was accomplished using two equivalents of NaNH_2 in liquid NH_3 . The more basic site in the dianion is selectively alkylated. Note that the dianion is an indenyl anion, and this may contribute to its accessibility by di-deprotonation.



- d. This is a nitrile alkylation involving an anion that is somewhat stabilized by conjugation with the indole ring. The anion was formed using NaNH_2 in liquid NH_3 .



- e. The structure of the anion is shown above. The anion is stabilized by resonance with the indole ring. The anion is formed using NaNH_2 in liquid NH_3 .
- f. The anion is shown above. The anion is stabilized by resonance with the indole ring. The anion is formed using NaNH_2 in liquid NH_3 .



Solution Manual For Advanced Organic Chemistry Part

E Durkheim



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