

SOLUBILITY CURVE WORKSHEET KEY

Use your solubility curve graphs provided to answer the following questions.

1. What are the customary units of solubility on solubility curves? Degrees Celsius and grams of solute/100g of water
2. Define solubility. A measure of how much solute can dissolve in a given amount of solvent.
3. According to the graph, the solubility of any substance changes as temperature changes.
4. List the substances whose solubility decreases as temperature increases. NaCl and CaSO_4
5. Which substance is least affected by temperature changes? NaCl
6. How many grams of ammonium chloride (NH_4Cl) at 50°C ? 50g
7. NaCl and KCl have the same solubility at approximately 75°C .
8. Which compound is least soluble in water at 10°C ? KClO_3
9. How many grams of KNO_3 can be dissolved at 50°C ? 80g
10. Are the following solutions unsaturated, saturated, or supersaturated?
 - a. 40g of NaNO_3 in 100 g of water at 30°C . saturated
 - b. 60g of KClO_3 in 100 g of water at 90°C . supersaturated
11. How many grams of sodium chloride, NaCl are required to saturate 100 grams of water at 100°C ? 40g
12. How many grams of NaNO_3 are required to saturate 100 grams of water at 75°C ? 140g
13. How many grams of KI will saturate water at 20°C ? 30g
14. At what temperature would 25g of potassium chlorate (KClO_3) dissolve? 85°C
15. At what temperature would 60g of NH_4Cl dissolve? 35°C
16. 80-g NaNO_3 is prepared at 30°C .

Solubility Answers Key

Kaplan Test Prep

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located on the right side of the slide.

Solubility Answers Key:

Regents Exams and Answers: Chemistry--Physical Setting Revised Edition Barron's Educational Series, Albert Tarendash, 2021-01-05 Barron's Regents Exams and Answers Chemistry provides essential practice for students taking the Chemistry Regents including actual recently administered exams and thorough answer explanations for all questions This book features Eight actual administered Regents Chemistry exams so students can get familiar with the test Thorough explanations for all answers Self analysis charts to help identify strengths and weaknesses Test taking techniques and strategies A detailed outline of all major topics tested on this exam A glossary of important terms to know for test day

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Assessment Ayman F. El-Kattan, 2017-05-15 Specifically geared to personnel in the pharmaceutical and biotechnology industries this book describes the basics and challenges of oral bioavailability one of the most significant hurdles in drug discovery and development Describes approaches to assess pharmacokinetics and how drug efflux and uptake transporters impact oral bioavailability Helps readers reduce the failure rate of drug candidates when transitioning from the bench to the clinic during development Explains how preclinical animal models used in preclinical testing and in vitro tools translate to humans which is an underappreciated and complicated area of drug development Includes chapters about pharmacokinetic modelling the Biopharmaceutics Drug Disposition Classification System BDDCS and the Extended Clearance Classification System ECCS Has tutorials for applying strategies to medicinal chemistry practices of drug discovery development MCAT General Chemistry Review 2025-2026 Kaplan Test Prep, 2024-08-13 Kaplan s MCAT General Chemistry Review 2025 2026

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