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#### Solubility Curve Practice Problems Worksheet 1

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You'll notice that for most substances, solubility increases as temperature increases. As discussed earlier in solutions involving liquids and solids typically more solute can be dissolved at higher temperatures. Can you find any

exceptions on the graph?\_\_\_\_\_

Here's an example of how to read the graph. Find the curve for KClOs.

At 30°C approximately 10g of KClO<sub>3</sub> will dissolve in 100g of water. If the temperature is increased to 80°C, approximately \_\_\_\_\_\_ of the substance will dissolve in 100g (or 100mL) of water.

Directions: Use the graph to answer the following questions. REMEMBER UNITS!

- What mass of solute will dissolve in 100mL of water at the following temperatures?
  - a. KNO3 at 70°C = \_\_\_\_\_\_
  - b. NaCl at 100°C= \_\_\_\_\_
  - c. NH<sub>4</sub>Cl at 90°C=
  - d. Which of the above three substances is most soluble in water at 15°C. =



# **Solubility Curve Practice Problems 1 Answer**

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