- (5 points) Table 4.8 (<u>Snoeyink</u> and Jenkins) shows the concentration of carbonate species for solutions with distilled water and 10-5M solutions of H₂CO₃*; sodium bicarbonate, NaHCO₃; and sodium carbonate, Na₂CO₃, in a system open to the atmosphere and in a closed system.
 - a. Why does the concentration of H₂CO₃* remain constant in the open system (across the different solutions) and decreases in a closed system?
 - b. For both systems, the concentration of H* decreases when comparing a solution of distilled water and 10-5M solutions of H2CO3*; NaHCO3 Na2CO3 with lowest concentration observed for Na2CO3 The decrease in H* concentration is greater for a closed system than for an open system. Explain this difference.

TABLE 4.8 Comparison of Composition and pH of Open and Closed Carbonate Systems

Solution	Closed System (Fig. 4-17)						
	[H*]	[OH-]	[H _i CO;]	[HCO,]	[CO,1-]	pН	C1,00
10-4 M H,CO;	10-4.7	10-4.9	10-41	10-4.7	10-16.4	5.7	10-54
10" M NaHCO,	10-7.4	10-4.4	10-44	10-1	10-54	7.6	10-5.0
10-4 M Na,CO,	10-*	10-	10-7.4	10-6	10-4.0	9.0	10-44
Solution	Open System						
	[H*]	[OH-]	[H,CO;]	[HCO,-]	(CO,*-)	pН	C1.00,
10-4 M H,CO;	10-6.7	10-43	10-4	10-4.7	10-16.4	5.7	10-4.0
10" M NaHCO,	10-4.3	10 1.7	10-	10-4	10-44	6.3	10-4
10-4 M Na,00,	10-43	10-7.3	10-*	10-47	10 **	6.6	10-44

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presenting recent research advances and applications of technologies in this important and challenging area of engineering. It highlights full scale applications to illustrate the performance and effectiveness of the new technologies. As an essential element for all living organisms P cannot be replaced by any other element in biochemical processes humans ultimately rely its availability. Today P is mostly obtained from mined rock phosphate Pi However natural reserves of high grade rock Pi are limited and dwindling on a global scale. As such there have been increased efforts to recycle P from secondary sources including sewage sludge animal manure food waste and steelmaking slag and so close the anthropogenic P cycle. In addition to various aspects of phosphorus covered by other literature including chemistry biochemistry ecology soil plant systems and sustainable management this book is a valuable and comprehensive source of information on the rapidly evolving field of P recovery and recycling engineering for students researchers and professionals responsible for sustainable use of phosphorus

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