

SOLUBILITY CURVE WORKSHEET KEY

Use your solubility curve graphs provided to answer the following questions.

1. What are the customary units of solubility on solubility curves? Degrees Celsius and grams of solute/100g of water
2. Define solubility. A measure of how much solute can dissolve in a given amount of solvent.
3. According to the graph, the solubility of any substance changes as temperature changes.
4. List the substances whose solubility decreases as temperature increases. NaCl and CaSO_4
5. Which substance is least affected by temperature changes? NaCl
6. How many grams of ammonium chloride (NH_4Cl) at 50°C ? 50g
7. NaCl and KCl have the same solubility at approximately 75°C .
8. Which compound is least soluble in water at 10°C ? KClO_3
9. How many grams of KNO_3 can be dissolved at 50°C ? 80g
10. Are the following solutions unsaturated, saturated, or supersaturated?
 - a. 40g of NaNO_3 in 100 g of water at 30°C . saturated
 - b. 60g of KClO_3 in 100 g of water at 90°C . supersaturated
11. How many grams of sodium chloride, NaCl are required to saturate 100 grams of water at 100°C ? 40g
12. How many grams of NaNO_3 are required to saturate 100 grams of water at 75°C ? 140g
13. How many grams of KI will saturate water at 20°C ? 30g
14. At what temperature would 25g of potassium chlorate (KClO_3) dissolve? 85°C
15. At what temperature would 60g of NH_4Cl dissolve? 35°C
16. 80-g NaNO_3 is prepared at 30°C .

Solubility Packet Answers

DJ Losen



Solubility Packet Answers:

Unveiling the Magic of Words: A Overview of "**Solubility Packet Answers**"

In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is actually awe-inspiring. Enter the realm of "**Solubility Packet Answers**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

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