

SOLUBILITY CURVES

Name _____

Answer the following questions based on the solubility curve below.

- Which salt is least soluble in water at 20° C? KClO₃
- How many grams of potassium chloride can be dissolved in 200 g of water at 80° C?
100 g
- At 40° C, how much potassium nitrate can be dissolved in 300 g of water? 123 g
- Which salt shows the least change in solubility from 0° - 100° C?
NaCl
- At 30° C, 90 g of sodium nitrate is dissolved in 100 g of water. Is this solution saturated, unsaturated or supersaturated?
unsaturated
- A saturated solution of potassium chlorate is formed from one hundred grams of water. If the saturated solution is cooled from 80° C to 50° C, how many grams of precipitate are formed? 40 - 20 g = 20 g
- What compound shows a decrease in solubility from 0° to 100° C? NH₃
- Which salt is most soluble at 10° C? KI
- Which salt is least soluble at 50° C? KClO₃
- Which salt is least soluble at 90° C? NH₃



Solubility Curve Practice Key

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A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the Manhattan Prep logo.

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Developing Solid Oral Dosage Forms Yihong Qiu, Yisheng Chen, Geoff G.Z. Zhang, Lawrence Yu, Rao V.

Mantri, 2016-11-08. *Developing Solid Oral Dosage Forms: Pharmaceutical Theory and Practice*, Second Edition, illustrates how to develop high-quality, safe, and effective pharmaceutical products by discussing the latest techniques, tools, and scientific advances in preformulation, investigation, formulation, process design, characterization, scale-up, and production operations. This book covers the essential principles of physical pharmacy, biopharmaceutics, and industrial pharmacy and their application to the research and development process of oral dosage forms. Chapters have been added, combined, deleted, and completely revised as necessary to produce a comprehensive, well-organized, valuable reference for industry professionals and academics engaged in all aspects of the development process. New and important topics include spray drying, amorphous solid dispersion, using hot melt extrusion, modeling and simulation, bioequivalence of complex modified-release dosage forms, biowaivers, and much more. Written and edited by an international team of leading experts with experience and knowledge across industry, academia, and regulatory settings. Includes new chapters covering the pharmaceutical applications of surface phenomenon, predictive biopharmaceutics, and pharmacokinetics; the development of formulations for drug discovery support; and much more. Presents new case studies throughout and a section completely devoted to regulatory aspects, including

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