

Solubility and Concentration

1. Match the following. Write the correct letter in the blank provided.

1. Solute <u>B</u>	A. Tells you how much solvent is dissolved in the solution
2. Solvent <u>E</u>	B. The part that dissolves into the solution
3. Dissolve <u>D</u>	C. Increase solubility by moving molecules faster
4. Stirring <u>C</u>	D. When something is seen to disappear inside the solvent
5. Concentration <u>A</u>	E. The part that dissolves something

2. Determine if the following statements are true or false.

- a. Solubility increases if you stir a solute into a solution. F
 b. All solutes have the same solubility in a given solvent. F
 c. There is a limit on the amount of solute that can dissolve in a given solvent. T
 d. You can dissolve additional solutes in an unsaturated solution. T

3. Give three ways to make sugar dissolve faster. Heat Stirring and increasing surface area

4. Which one will dissolve faster? Circle the correct answer.

- a. Powdered sugar or granulated sugar? b. In hot water or in cold water?
 c. Stirred or not stirred? d. Large particles or small particles ?

5. 65 g of sugar is dissolved in 750 ml of water. What is the concentration of the solution?
0.08 g/ml

6. Which is more concentrated: 34 g of salt dissolved in 100 ml of water or 100 g of salt in 1500 ml of water?
34 g of Salt in 100ml of water

7. If the solubility of salt in water was determined to be 0.5 g/ml, would a solution with 50 g of salt in 150 ml of water be considered saturated?

no

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J Spring



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