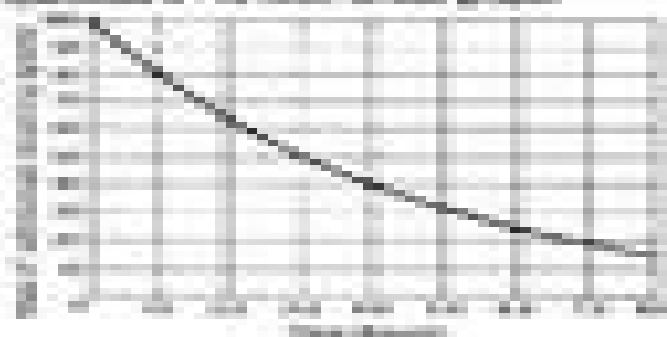


2-4- Nuclear Chemistry

AP EXAM OBJECTIVES

1. When Marguerite was born, her father (Mr. A.K.A. "Radioactive Father") was emitting alpha radiation.
(10% emitting the most "radioactivity")
(5% emitting alpha radiation)
(5% emitting beta radiation)
2. When Marguerite's father was born, his father (Mr. A.K.A. "Radioactive Grandfather") was emitting alpha radiation.
(10% emitting alpha radiation)
(5% emitting beta radiation)
(5% emitting gamma radiation)
3. Radioactive isotopes are "radioactive". They emit other particles or photons.
(A.K.A. Ray 1.00) (A.K.A. Ray 2.00)
(10% Ray 1.00) (10% Ray 2.00)
4. What is the charge carried by alpha particles?
(A.K.A. $+2$) (10% $+1$) (10% $+3$) (10% -1)
5. What type of radiation is alpha or beta radiation? Name it right!
(A.K.A. alpha) (10% gamma)
(10% beta) (10% gamma)
6. What did the son spontaneously emit a half-life of 1.5 hours. What was it the following substance is then after 3.0 hours after 1.5 hours?
(A.K.A. half of the half-life is then emitted)
(10% 75% of the half-life is then emitted)
(10% 100% of the half-life is then emitted)
(10% 25% of the half-life is then emitted)
7. In a decay series of the (123) the first three steps involve alpha decay and then beta decay. What is the result of alpha decay?
(A.K.A. $2\alpha - 4\alpha$) (10% $2\alpha - 4\alpha$)
(10% $2\alpha - 4\alpha$) (10% $2\alpha - 4\alpha$)

Questions 8 - 10 refer to the graph.



8. According to the above data, what is the half-life of the substance?
(10% 1.5 hours) (10% 1.5 hours)
(10% 1.5 hours) (10% 1.5 hours)
9. What percent of the original sample remains after 3 hours?
(10% 100%) (10% 100%)
(10% 75%) (10% 100%)
10. What is the graph above a graph for a substance with half-life of 1.5 hours.
11. Radioactive 1.00 half-life of 1.5 hours. What percent of a sample remains after 3.0 hours?
(10% 100%) (10% 100%) (10% 100%) (10% 1.5 hours)
12. Which of the following describes what occurs in the chemical process?
(10%) is being emitted in fragmented form.
(10%) is emitted in alpha form a positive and an electron.
(10%) most light nuclei are composed from 2 protons and 2 neutrons.
(10%) is positive and negative particles from completely ionized

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