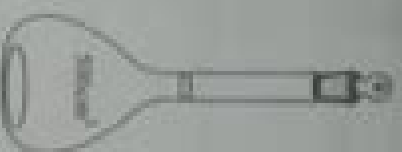


- 8.3 A learner prepares a standard solution by dissolving 5.3g of sodium carbonate, Na_2CO_3 in 500cm^3 of distilled water using the apparatus shown below.



- 8.3.1 Write down the name of the apparatus shown above.

(1)

- 8.3.2 Calculate the concentration of the sodium carbonate solution.

(4)

- 8.4 The learner used standard solution prepared in 8.3 to determine the concentration of a sample of hydrochloric acid (HCl). 25cm^3 of the sodium carbonate solution and added a few drops of phenolphthalein indicator. $18,65\text{cm}^3$ of the Hydrochloric acid reacted completely with the sodium carbonate. The reaction that takes place is:



- 8.4.1 What is the colour change at endpoint?

(2)

- 8.4.2 Calculate the concentration of the hydrochloric acid sample.

(3)

[24]

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Table of Contents Smart Learner Question Papers Physical Sciences

1. Understanding the eBook Smart Learner Question Papers Physical Sciences
 - The Rise of Digital Reading Smart Learner Question Papers Physical Sciences
 - Advantages of eBooks Over Traditional Books
2. Identifying Smart Learner Question Papers Physical Sciences
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Smart Learner Question Papers Physical Sciences
 - User-Friendly Interface
4. Exploring eBook Recommendations from Smart Learner Question Papers Physical Sciences
 - Personalized Recommendations

- Smart Learner Question Papers Physical Sciences User Reviews and Ratings
- Smart Learner Question Papers Physical Sciences and Bestseller Lists
- 5. Accessing Smart Learner Question Papers Physical Sciences Free and Paid eBooks
 - Smart Learner Question Papers Physical Sciences Public Domain eBooks
 - Smart Learner Question Papers Physical Sciences eBook Subscription Services
 - Smart Learner Question Papers Physical Sciences Budget-Friendly Options
- 6. Navigating Smart Learner Question Papers Physical Sciences eBook Formats
 - ePub, PDF, MOBI, and More
 - Smart Learner Question Papers Physical Sciences Compatibility with Devices
 - Smart Learner Question Papers Physical Sciences Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Smart Learner Question Papers Physical Sciences
 - Highlighting and Note-Taking Smart Learner Question Papers Physical Sciences
 - Interactive Elements Smart Learner Question Papers Physical Sciences
- 8. Staying Engaged with Smart Learner Question Papers Physical Sciences
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Smart Learner Question Papers Physical Sciences
- 9. Balancing eBooks and Physical Books Smart Learner Question Papers Physical Sciences
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Smart Learner Question Papers Physical Sciences
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Smart Learner Question Papers Physical Sciences
 - Setting Reading Goals Smart Learner Question Papers Physical Sciences
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Smart Learner Question Papers Physical Sciences
 - Fact-Checking eBook Content of Smart Learner Question Papers Physical Sciences

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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