



**PHYSICAL SCIENCES (CHEMISTRY)
GRADE 11 TERM 3 FORMAL EXPERIMENT - PART 2**

TIME: 1 HOUR

**VERIFICATION OF BOYLE'S LAW
INDIVIDUAL WORK (should be done under strict supervision).**

1. Write a scientific report which must include the following:

1.1 Aim

(1)

1.2 Apparatus and method

(5)

2. Observations and Recording of results (Copy the table below onto your book or sheet of paper and transfer the results you recorded when conducting the experiment). Calculate pV for each set of results and record your answers in the table.

Volume (V) cm^3	Pressure (p) kPa		pV (J)

(6)

3. Analysis of the results:

3.1 Identify the following variables for the experiment:

- Independent
- Dependent
- Controlled

(3)

3.2 Use the graph paper provided by your teacher to draw graphs of V versus p and V versus $\frac{1}{p}$.

(18)

4. Conclusion

4.1 Examine the results in your table as well as the two graphs you have drawn, what happens to the volume of the gas in the tube as the pressure is decreased.

(1)

4.2 Write down a conclusion for this experiment.

(3)

5

Questions and discussion:

5.1 Explain why you have to wait for a few minutes before you can take readings.

(2)

5.2 Explain the relationship between pressure and volume in terms of the kinetic theory of gases.

(4)

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WJ Hussar



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Table of Contents Smart Question Paper Physical Science Grade 11 P1

1. Understanding the eBook Smart Question Paper Physical Science Grade 11 P1
 - The Rise of Digital Reading Smart Question Paper Physical Science Grade 11 P1
 - Advantages of eBooks Over Traditional Books
2. Identifying Smart Question Paper Physical Science Grade 11 P1
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Smart Question Paper Physical Science Grade 11 P1
 - User-Friendly Interface
4. Exploring eBook Recommendations from Smart Question Paper Physical Science Grade 11 P1
 - Personalized Recommendations
 - Smart Question Paper Physical Science Grade 11 P1 User Reviews and Ratings
 - Smart Question Paper Physical Science Grade 11 P1 and Bestseller Lists

5. Accessing Smart Question Paper Physical Science Grade 11 P1 Free and Paid eBooks
 - Smart Question Paper Physical Science Grade 11 P1 Public Domain eBooks
 - Smart Question Paper Physical Science Grade 11 P1 eBook Subscription Services
 - Smart Question Paper Physical Science Grade 11 P1 Budget-Friendly Options
6. Navigating Smart Question Paper Physical Science Grade 11 P1 eBook Formats
 - ePub, PDF, MOBI, and More
 - Smart Question Paper Physical Science Grade 11 P1 Compatibility with Devices
 - Smart Question Paper Physical Science Grade 11 P1 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Smart Question Paper Physical Science Grade 11 P1
 - Highlighting and Note-Taking Smart Question Paper Physical Science Grade 11 P1
 - Interactive Elements Smart Question Paper Physical Science Grade 11 P1
8. Staying Engaged with Smart Question Paper Physical Science Grade 11 P1
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Smart Question Paper Physical Science Grade 11 P1
9. Balancing eBooks and Physical Books Smart Question Paper Physical Science Grade 11 P1
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Smart Question Paper Physical Science Grade 11 P1
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Smart Question Paper Physical Science Grade 11 P1
 - Setting Reading Goals Smart Question Paper Physical Science Grade 11 P1
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Smart Question Paper Physical Science Grade 11 P1
 - Fact-Checking eBook Content of Smart Question Paper Physical Science Grade 11 P1
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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