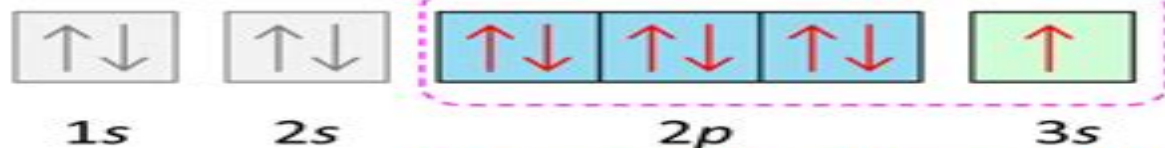


Electron Configuration of Main Group Cations

Electron configuration

Na $1s^2 2s^2 2p^6 3s^1$

Orbital diagram

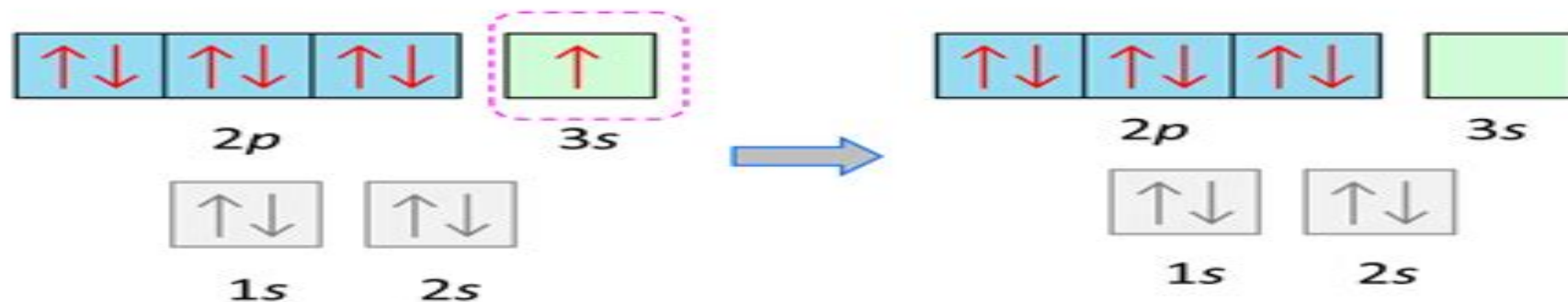


Focus on the valence shell

The charge is +1, therefore, one electron is lost:



The electron must be from the outermost valence orbital – $3s^1$



Na $1s^2 2s^2 2p^6 3s^1$

Na⁺ $1s^2 2s^2 2p^6$

↓ Same as

Ne $1s^2 2s^2 2p^6$



ChemistrySteps.com

Cations of main group metals attain the electron configuration of the noble gas that precedes them in the periodic table. They are **isoelectronic** with the previous noble gas.

Small Scale Lab Electron Configurations Of Ions

Peipei Pang



Small Scale Lab Electron Configurations Of Ions:

Whispering the Techniques of Language: An Emotional Quest through **Small Scale Lab Electoron Configurations Of Ions**

In a digitally-driven world wherever monitors reign great and immediate transmission drowns out the subtleties of language, the profound secrets and psychological nuances concealed within words usually get unheard. However, located within the pages of **Small Scale Lab Electoron Configurations Of Ions** a captivating fictional treasure blinking with organic thoughts, lies an exceptional journey waiting to be undertaken. Composed by an experienced wordsmith, that marvelous opus attracts readers on an introspective trip, lightly unraveling the veiled truths and profound impact resonating within the material of each word. Within the emotional depths of this emotional evaluation, we shall embark upon a genuine exploration of the book is core themes, dissect its captivating writing type, and yield to the effective resonance it evokes heavy within the recesses of readers hearts.

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