

Answer key

1 Building a career

1.1 About business The education business

1 telemarketing, personal referrals, weekends at Harvard, mail shots, travelling shows

- 2**
- 23,000 applications for the new academic year
 - 9% of applications were accepted
 - 80% ultimately decided to attend
 - 10% of admitted students came from abroad
 - 35 staff in Fitzsimmons' team
 - \$43,700 – the annual tab for tuition, room and board
 - \$60,000 – parents who make less aren't expected to pay
 - 8000 alumni volunteers

- 3**
- Purchase high school juniors' names from examination boards.
 - Send 70,000 or so letters to high school juniors.
 - Visit 140 US cities and travel to other countries.
 - Send out 8,000 alumni volunteers to identify possible candidates.
 - Sort applications and give a score from one to six.
 - Local subcommittees discuss the case for and against each candidate.
 - Professors, alumni, and students lobby successful candidates by phone.
 - Invite applicants to spend a weekend at Harvard in April.

- 4**
- speaker 3
 - speaker 1
 - speaker 4
 - speaker 2

1.2 Vocabulary Education and career

- 1** gain 2 acquire 3 improving 4 boosting 5 have
6 going to 7 making 8 receive 9 obtain 10 enhance

2 I worked in several departments and gained knowledge about all aspects of the business.
I believe this job will allow me to acquire new skills and boost my earnings.
My gap year in the States definitely improved my employability, and working for IBM certainly enhanced my CV.
On my Business Studies course I received tuition from practising managers and obtained experience on work placements.

- 3**
- finance
 - arrange
 - borrow
 - support
 - obtain
 - win
 - subsidise

4 Student A:
I wasn't a very motivated student when I was younger, so I failed my exams in 1999 and had to retake them the following summer. I'd already decided I wanted to take a gap year then go to business school, so this time I completed all the coursework and revised thoroughly; I passed easily. I spent a year abroad as an assistant in a secondary school and then went to business school – I'd already applied and successfully attended an interview before revisiting my exams.
In my second year as an undergraduate, I spent a semester as an intern at Hewlett Packard, which was very rewarding. In fact, I went straight into a job at HP after I graduated the following June. They encouraged me to do a Master's degree by correspondence – I just have to finish writing a dissertation. I need a break now, but I may take a sabbatical to study for an MBA in a few years' time.

Student B:
I wasn't a very motivated student when I was younger, so I had to retake my exams in 2000 after I failed them the year before. I'd already decided that before going to business school I wanted to take a gap year. I passed easily because this time I completed all the coursework and revised thoroughly. I'd already applied and successfully attended an interview at a business school, so after revisiting my exams I spent a year abroad as an assistant in a secondary school.
I spent a semester as an intern at Hewlett Packard in my second year as an undergraduate, which was very rewarding. In fact, after I graduated the following June I went straight into a job at HP. I'm now writing a dissertation for a Master's degree by correspondence which HP encouraged me to do. I may take a sabbatical to study for an MBA in a few years' time, but I need a break first.

- 5**
- contacted
 - presented
 - recruited
 - trained
 - met
 - succeeded
 - designed

- 6** implemented
7 managed / motivated
8 motivated / managed
9 chained
10 drew up

- 11**
- to be resting, to take a sabbatical
 - to be laid off, to be made redundant
 - to be promoted, to join a company
 - to quit, to give in your notice
 - to be on assignment, to be transferred
 - to be suspended, to be dismissed

- 12**
- He's intending to quit his job because his company keeps giving him assignments in different countries. He needs to give in his notice three months before he can leave.
 - She was laid off when they modernised the company, then made redundant when they couldn't find any work for her.
 - He's been suggested for not wearing a safety helmet, but he thinks the company want to dismiss him.
 - He joined the company four years ago and has just been promoted.
 - She's on secondment in Paris, and she's being transferred to Vancouver.
 - He's resting at the moment, and considering taking a sabbatical to write a book.

Name: Bob	Name: Lucy	Name: Mel
Problem: Started up successful electronics company. Resigned when he disagreed with his CEO. Sell shares? What next?	Problem: Previously worked as assistant teacher. Promoted when she wrote stories at work. Do creative writing course, or look for another job?	Problem: Designed and sold product to local govt. When in high school. Drop out of prestigious business school to start own business?

1.3 Grammar Tense review

- 1** When I left school I wasn't sure what to do next. I had applied for a place at university, but while I was revising for my exams I had already decided that I didn't feel ready for university. I had read a lot about East Africa, and one day when I was watching a TV documentary about Ethiopia, I suddenly knew that that was where I was going. In fact it was while I was working with an NGO in Ethiopia that I became interested in business.

- 2**
- have just finished
 - am currently working
 - consider
 - has left
 - have agreed
 - have been designing
 - have been looking
 - seems

- 3**
- If I get the job, I will try to make myself available as soon as possible, but I wouldn't be able to start immediately unless HPC agreed to let me go.
 - You see, if I make a commitment, I do my best to respect it.
 - If by any chance the job were based in Kenya, I would take it anyway, even though I'd prefer to use my Ambulic.

4 In the short term, I hope (1) I'll become a department manager quite quickly. We learnt a lot about management on the MBA, and (2) I'm going to put the theory into practice as soon as I can. On the personal front, (3) I'm getting married next June; over the next couple of years my fiancé (4) is going to write TV and movie scripts, so (5) he is going to be totally mobile and happy to move as necessary.

- 5** Exercise 5
- | | |
|---------------|--------------------|
| usually | present simple |
| currently | present continuous |
| always | present simple |
| at present | present continuous |
| at the moment | present continuous |
| often | present simple |

- 6** Exercise 6
- | | |
|-----------|-----------------|
| ever | present perfect |
| in | past simple |
| never | present perfect |
| ago | past simple |
| already | present perfect |
| (not) yet | present perfect |
| when | past simple |
| just | present perfect |

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Carol A. Chapelle, Shannon Sauro



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web the answer is rather simple if you understand electron configurations the shape of the periodic table mimics the filling of the subshells with electrons let us start with h and he their electron configurations are $1s^1$ and $1s^2$ respectively with he the n 1 shell is filled

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web jun 14 2015 this list of electron configurations of elements contains all the elements in increasing order of atomic number to save room the configurations are in noble gas shorthand this means part of the electron configuration has been replaced with the element symbol of the noble gas symbol

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web elements in the second row of the periodic table place their electrons in the 2n shell as well as the 1n shell for instance lithium li has three electrons two fill the 1 s orbital and the third is placed in the 2 s orbital giving an electron configuration of 1 s 2 2 s 1

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web one more point needs to be emphasized about the relationship between electron configuration and the periodic table the atoms of elements in the same vertical column of the table have similar electron configurations for example consider the alkaline earth elements group iia

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web the electron removed during the ionization of beryllium he 2s 2 2 is an s electron whereas the electron removed during the ionization of boron he 2s 2 2p 1 is a p electron this results in a lower first ionization energy for boron even though its nuclear charge is greater by one proton

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web predictions from reliable sources have been used for these elements grayed out electron numbers indicate subshells filled to their maximum bracketed noble gas symbols on the left represent inner configurations that are the same in each

period written out these are he 2 helium 1s 2 ne 10 neon 1s 2 2s 2 2p 6

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web q1 what is the electron configuration of the element potassium answer the atomic number of potassium is 19 a neutral potassium element has 19 electrons its electron configuration is 1s 2 2s 2 2p 6 3s 2 3p 6 4s 1 or simply put as 4s 1 q2 what is the total number of electrons that can be filled in s p and d subshells

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web by building up from hydrogen this table can be used to determine the electron configuration for any atom on the periodic table we will now construct the ground state electron configuration and orbital diagram for a selection of atoms in the first and second periods of the periodic table

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web electron configurations are a simple way of writing down the locations of all of the electrons in an atom as we know the positively charged protons in the nucleus of an atom tend to attract negatively charged electrons

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