

CHEMISTRY PAPER 1 (THEORY)

Maximum Marks: 70

Time Allowed: Three hours

*Candidates are allowed additional 15 minutes for only reading the paper.
They must NOT start writing during this time.*

This paper is divided into four sections – A, B, C and D.

Answer all questions.

Section A consists of one question having sub-parts of one mark each.

Section B consists of ten questions of two marks each.

Section C consists of seven questions of three marks each, and

Section D consists of three questions of five marks each.

Internal choices have been provided in one question each in Section B,

Section C and Section D.

All working, including rough work, should be done on the same sheet as, and adjacent to the rest of the answer.

The intended marks for questions or parts of questions are given in brackets [].

Balanced equations must be given wherever possible and diagrams where they are helpful.

When solving numerical problems, all essential working must be shown.

In working out problems, use the following data:

$$\begin{aligned}\text{Gas constant } R &= 1.987 \text{ cal deg}^{-1} \text{ mol}^{-1} = 8.314 \text{ JK}^{-1} \text{ mol}^{-1} \\ &= 0.0821 \text{ dm}^3 \text{ atm K}^{-1} \text{ mol}^{-1}\end{aligned}$$

$$1 \text{ l atm} = 1 \text{ dm}^3 \text{ atm} = 101.3 \text{ J, } 1 \text{ Faraday} = 96500 \text{ coulombs.}$$

$$\text{Avogadro's number} = 6.023 \times 10^{23}$$

SECTION A – 14 MARKS

Question 1

- (A) Fill in the blanks by choosing the appropriate word(s) from those given in the [4×1]
brackets:

[stable, low, aldehyde, unstable, 6, 4, ethane, Clemmensen's, 2, 3, carboxylic acid, high, propane, Rosenmund's]

- (i) The primary alcohols are easily oxidised first into _____ and then into _____.
- (ii) The intermediate activated complex in a chemical reaction is highly _____ due to _____ energy.
- (iii) The coordination number and oxidation state of the complex $K_4[Fe(CN)_6]$ are _____ and _____ respectively.
- (iv) Propanone on reaction with zinc-amalgam in presence of conc. HCl gives _____ and the reaction is known as _____ reduction.

This Paper consists of 8 printed pages.

Year 11 Chemistry Past Papers Final Exam

Educart, Dr. Rakshita Singh

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the text 'Educart, Dr. Rakshita Singh'.

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Table of Contents Year 11 Chemistry Past Papers Final Exam

1. Understanding the eBook Year 11 Chemistry Past Papers Final Exam
 - The Rise of Digital Reading Year 11 Chemistry Past Papers Final Exam
 - Advantages of eBooks Over Traditional Books
2. Identifying Year 11 Chemistry Past Papers Final Exam
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Year 11 Chemistry Past Papers Final Exam
 - User-Friendly Interface
4. Exploring eBook Recommendations from Year 11 Chemistry Past Papers Final Exam
 - Personalized Recommendations
 - Year 11 Chemistry Past Papers Final Exam User Reviews and Ratings
 - Year 11 Chemistry Past Papers Final Exam and Bestseller Lists

5. Accessing Year 11 Chemistry Past Papers Final Exam Free and Paid eBooks
 - Year 11 Chemistry Past Papers Final Exam Public Domain eBooks
 - Year 11 Chemistry Past Papers Final Exam eBook Subscription Services
 - Year 11 Chemistry Past Papers Final Exam Budget-Friendly Options
6. Navigating Year 11 Chemistry Past Papers Final Exam eBook Formats
 - ePub, PDF, MOBI, and More
 - Year 11 Chemistry Past Papers Final Exam Compatibility with Devices
 - Year 11 Chemistry Past Papers Final Exam Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Year 11 Chemistry Past Papers Final Exam
 - Highlighting and Note-Taking Year 11 Chemistry Past Papers Final Exam
 - Interactive Elements Year 11 Chemistry Past Papers Final Exam
8. Staying Engaged with Year 11 Chemistry Past Papers Final Exam
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Year 11 Chemistry Past Papers Final Exam
9. Balancing eBooks and Physical Books Year 11 Chemistry Past Papers Final Exam
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Year 11 Chemistry Past Papers Final Exam
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Year 11 Chemistry Past Papers Final Exam
 - Setting Reading Goals Year 11 Chemistry Past Papers Final Exam
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Year 11 Chemistry Past Papers Final Exam
 - Fact-Checking eBook Content of Year 11 Chemistry Past Papers Final Exam
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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