1) at What is a base? b) What is an alkali? c) What is the reaction between an acid and a base called? 67.5 d) Give two everyday examples of reactions between acids and bases. 623 a) Match the descriptions below to the pH's in this list: (20) description mentional weary advistings sliightly acidle. slightly alkaline wery acidio 694 b) i) Litmus indicator could not distinguish between solutions of pH 2 and 5. Explain why. 671.5 ii) Name an indicator that could. 28.6 Complete the following word equations: a) zinc + nitric acid -b) hydrochloric acid + copper carbonate → sulfuric acid + sodium oxide → d) potassium hydroxide + suffuric acid → e) hydrochloric acid + lithium carbonate → magnesium oxide + nitric acid -g) sulfuric acid + strontium --h) potassium carbonate + suffuric acid → rubidium hydroxide + hydrochloric scid → a) Why should you not add an alkali to an acid spill? b) Give two safer ways of treating acids spills.

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ML Yell

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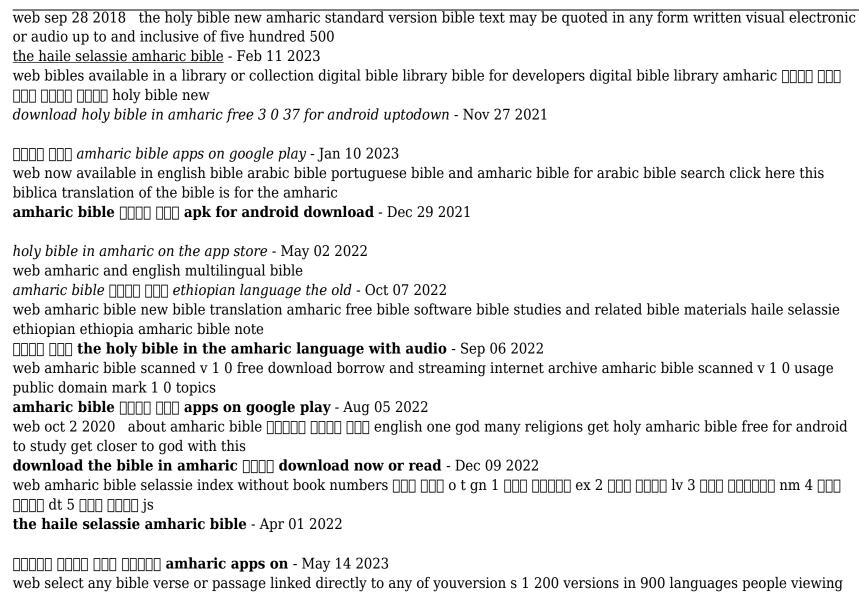
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