

ELECTRONICS

SEMICONDUCTOR

There are a large number of materials which have resistivities lying between those of an insulator and a conductor (See table). Such materials are known as 'semiconductors'.

At absolute zero, pure and perfect crystals of the semiconductors are nonconducting, their resistivity approaching to the resistivity of an insulator. They can be made conducting by adding impurities, thermal agitation and lattice defects etc.

Resistivity of a semiconductor depends upon the temperature, and it decreases with the rise in temperature; consequently, a semiconductor crystal becomes conducting even at room temperature.

At room temperature, their resistivity lies in the range 10^2 to 10^9 ohm-cm, and is thus intermediate between the resistivity of good conductor (10^{-8} ohm-cm) and insulator (10^{14} to 10^{22} ohm-cm.).

Electrical resistivity of various material at 20°C in ohm-meter.

Metals	Semiconductors	Insulators
Ag – 1.6×10^{-8}	Ge (pure) – 0.47	glass – 10^{10} – 10^{11}
Cu – 1.7×10^{-8}	Si (pure) – 1953	mica – 7×10^{14}
Al – 2.8×10^{-8}	Fe_3O_4 – 0.01	diamond – 10^{14}
Fe – 10×10^{-8}	In Sb – 2×10^4	
Constantan – 49×10^{-8}	Sn (gray) – 2×10^5	
Nichrome – 100×10^{-8}		

At 20°C resistivity of semiconductors lying between metals and insulators. But at low temperatures, semiconductor behaves as insulator, because resistivity of semiconductor depends strongly on temperature.

Properties :

- They have negative temperature coefficient of resistance means resistance of semiconductor decreases with increase in temperature and vice-versa.
- Their electrical conductivity is very much affected by even a very minute amount of impurity added to it.
- Covalent Bond
- Crystalline Structure

Silicon ($_{14}\text{Si}^{28}$), germanium ($_{32}\text{Ge}^{72}$) and tin ($_{50}\text{Sn}^{118}$) in crystalline form have the same crystal structure and similar electrical properties.

Silicon and germanium are most popular semiconductors, because of the importance of Ge and Si in present day electronics.

There are also a number of compound semiconductors, such as metallic oxides and sulphides which are also of great importance.

Ex. GaAs and PbSe and ternary alloys are of very importance in present day electronics.

Elements enclosed by the line are semiconductors

III A	IV A	V A	VI A	VII A
B	C	N	O	F
Al	Si	P	S	Cl
Ga	Ge	As	Se	Br
In	Sn	Sb	Te	I
Tl	Pb	Bi	Po	At

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Henry Stoneley Burhop, H. B. Gilbody**

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Table of Contents Target 12th Electronics Notes

1. Understanding the eBook Target 12th Electronics Notes
 - The Rise of Digital Reading Target 12th Electronics Notes
 - Advantages of eBooks Over Traditional Books
2. Identifying Target 12th Electronics Notes
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Target 12th Electronics Notes
 - User-Friendly Interface
4. Exploring eBook Recommendations from Target 12th Electronics Notes
 - Personalized Recommendations
 - Target 12th Electronics Notes User Reviews and Ratings
 - Target 12th Electronics Notes and Bestseller Lists
5. Accessing Target 12th Electronics Notes Free and Paid eBooks

- Target 12th Electronics Notes Public Domain eBooks
- Target 12th Electronics Notes eBook Subscription Services
- Target 12th Electronics Notes Budget-Friendly Options
- 6. Navigating Target 12th Electronics Notes eBook Formats
 - ePub, PDF, MOBI, and More
 - Target 12th Electronics Notes Compatibility with Devices
 - Target 12th Electronics Notes Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Target 12th Electronics Notes
 - Highlighting and Note-Taking Target 12th Electronics Notes
 - Interactive Elements Target 12th Electronics Notes
- 8. Staying Engaged with Target 12th Electronics Notes
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Target 12th Electronics Notes
- 9. Balancing eBooks and Physical Books Target 12th Electronics Notes
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Target 12th Electronics Notes
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Target 12th Electronics Notes
 - Setting Reading Goals Target 12th Electronics Notes
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Target 12th Electronics Notes
 - Fact-Checking eBook Content of Target 12th Electronics Notes
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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