#### Chemistry Test - Nuclear Chemistry

| A. | Balancing   | Nuclear Reactions.  | Solve t | tio. | following  | nuclear reactions and | WITH |
|----|-------------|---------------------|---------|------|------------|-----------------------|------|
|    | the correct | answer in the large | blank.  | -4   | pts. each. |                       |      |

1. ""ωPo → "He + \_\_\_

2. "<sub>s</sub>, P → '<sub>s</sub>e + \_\_\_\_

3. ",C + ",y + \_\_\_

5. \*\*\*...U + 3 \*.n +

6 \*\*\*...Pb + \*\*n +

7. ™<sub>v</sub>Am → ',He + ',v +

8. \*\*\*<sub>\*\*</sub>Bi → \*\*\*<sub>\*\*</sub>Tl +

9 \*\*\* AU - I\*\* Ha +

10 \*\*\*-Pt + \*\*\*-Pt +

B. True and False. Identify the following statements as true or false. CORRECT FALSE STATEMENTS FOR BONUS POINTSHI 2 pts each.

\_\_\_\_\_1. Fusion is the splitting of a large, unstable nucleus.

2. Fusion occurs on the sun and stars.

Roentgen discovered radioactivity.

4. Marie Cure discovered x-rays.

 In a nuclear power plant, heat from a nuclear reaction is used to generate electricity.

The largest radioactive particle is the alpha particle.

Alpha particles are the most dangerous because they easily enter the cell.

Gamma rays are actually tiny electron-like particles.

Nuclear reactions involve the nucleus of the atom.

Nuclear reactions can create new elements, unlike chemical reactions.

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Zhe Xu

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