

Chemistry VIVA Questions & Answers

1) Explain the term Aliquot and Diluent?

- Aliquot : It is a measured sub-volume of original sample
- Diluent: Material with which sample is diluted

2) Explain what is molality?

Molality is the number of solute that is present in 1 kg of a solvent.

3) Explain what is titration?

Titration is a process to determine the molarity of a base or an acid. In this process a reaction is carried out between the known volumes of a solution with a known concentration, against the known volume of a solution with an unknown concentration.

4) Explain what is buffer?

A buffer is an aqueous solution which has highly stable pH. It is a blend of a weak acid and its conjugate base or vice versa. On adding small amount of base or acid to buffer, its pH hardly changes.

5) Explain how buffer works?

In buffer when hydrogen ion is added, it will be neutralized by the base in buffer. Hydroxide ion will be neutralized by the acid. On the overall pH of the buffer solution, these neutralization reactions will not show much effect.

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Mark Steyvers



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