

Exercises 10 – Last Name – First Name – ID# The problem set has questions 1 – 10 and 12 questions 11 and 13.

Difficult part first:

1. Which of these is temperature (in the proper sense) rather than enthalpy? (1)
2. The standard enthalpy of formation of H_2O is (1)
3. The single factor which makes H_2O a more stable molecule is (1)
4. $\Delta_{\text{vap}}H^\circ$ for water is 40.7 kJ mol^{-1} . What is $\Delta_{\text{cond}}H^\circ$ for water? (1)
5. The standard $\Delta_{\text{f}}H^\circ$ for H_2O is $-285.8 \text{ kJ mol}^{-1}$. What is $\Delta_{\text{f}}H^\circ$ for H_2O at 100°C ? (1)
6. What is the $\Delta_{\text{f}}H^\circ$ for H_2O at 100°C ? (1)
7. The $\Delta_{\text{f}}H^\circ$ for H_2O is $-285.8 \text{ kJ mol}^{-1}$. What is $\Delta_{\text{f}}H^\circ$ for H_2O at 100°C ? (1)
8. $\Delta_{\text{f}}H^\circ$ for H_2O is $-285.8 \text{ kJ mol}^{-1}$. What is $\Delta_{\text{f}}H^\circ$ for H_2O at 100°C ? (1)

10. (1)

11. (1)

12. (1)

13. (1)

14. (1)

15. (1)

16. (1)

17. (1)

18. (1)

19. (1)

20. (1)

21. (1)

22. (1)

23. (1)

Vista Leccion 9 Lab Manual Answers

Jarvis, Ana C. Jarvis



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