

Build an Atom Virtual Lab

3 pages + 6-page Answer Key

Chemistry

Physical Science

BUILD AN ATOM VIRTUAL LAB

Introduction:

In this lab you will understand the structure of atoms, including protons, neutrons, and electrons, and how these particles determine the identity and properties of elements.

Materials:

Virtual Lab: Build an Atom (https://phet.colorado.edu/en/simulations/build-an-atom)

Procedure (Part 1)

1. Access the Lab: Enter type in the search bar.
2. Build at least 8 different atoms by adjusting the number of protons, neutrons, and electrons. Record the atomic number, mass number, and charge of each atom in the table below.
3. Observe how the number of protons, neutrons, and electrons affect the identity and properties of the atom.

Element	Symbol	Protons	Neutrons	Electrons	Charge

Atoms



UNDER CONSTRUCTION

When building an atom, write the information:

1. Atomic Number (Z): Number of protons in the nucleus.

2. Mass Number (A): Sum of protons and neutrons in the nucleus.

3. Charge: Difference between the number of protons and electrons.

4. Element Symbol: Chemical symbol for the element.

5. Isotope Name: Name of the specific isotope.

6. Atomic Weight: Average mass of the atoms.

7. Periodic Table: Location of the element.

8. Group Number: Number of the group in the periodic table.

9. Period Number: Number of the period in the periodic table.

10. Block: s-block, p-block, d-block, or f-block.

11. Valence Electrons: Electrons in the outermost shell.

12. Ionization Energy: Energy required to remove an electron.

13. Electronegativity: Ability to attract electrons.

14. Atomic Radius: Distance from the nucleus to the outermost electron.

15. Density: Mass per unit volume.

16. Melting Point: Temperature at which the solid turns into a liquid.

17. Boiling Point: Temperature at which the liquid turns into a gas.

18. Half-life: Time taken for half of the substance to decay.

19. Radioactivity: Emission of particles or energy from the nucleus.

20. Stability: Whether the atom is stable or radioactive.

21. Abundance: How common the element is in nature.

22. Uses: Applications of the element in industry or science.

23. Discovery: Who discovered the element and when.

24. Historical Significance: Role of the element in the development of science.

25. Future Prospects: Potential applications of the element in the future.

26. Environmental Impact: Effect of the element on the environment.

27. Health Effects: Effect of the element on human health.

28. Safety Precautions: Measures to take when handling the element.

29. Storage: How to store the element safely.

30. Disposal: How to dispose of the element safely.

31. Recycling: How to recycle the element.

32. Conservation: How to conserve the element.

33. Legislation: Laws governing the use of the element.

34. International Cooperation: Collaboration between countries in the study of the element.

35. Public Awareness: Efforts to educate the public about the element.

36. Funding: Sources of funding for research on the element.

37. Collaboration: Partnerships between academia, industry, and government.

38. Innovation: New technologies developed from research on the element.

39. Education: Integration of the element into school curricula.

40. Outreach: Activities to engage the public in the study of the element.

41. Policy Development: Guidelines for the use of the element.

42. Regulatory Framework: Rules governing the use of the element.

43. Standardization: Establishment of standards for the use of the element.

44. Quality Control: Measures to ensure the quality of the element.

45. Certification: Processes for certifying the purity of the element.

46. Accreditation: Recognition of the research on the element.

47. Peer Review: Evaluation of research on the element by other scientists.

48. Publication: Dissemination of research on the element to the scientific community.

49. Archiving: Preservation of research on the element for future generations.

50. Open Access: Making research on the element available to all.

51. Data Sharing: Exchange of data between researchers.

52. Reproducibility: Ensuring that research on the element can be repeated.

53. Transparency: Openness in the research process.

54. Accountability: Responsibility for the use of the element.

55. Ethics: Moral principles governing the use of the element.

56. Governance: Management of the element.

57. Oversight: Monitoring the use of the element.

58. Review: Evaluation of the use of the element.

59. Reporting: Communication of the use of the element.

60. Documentation: Recording the use of the element.

61. Archiving: Preservation of the use of the element.

62. Access: Making the use of the element available.

63. Sharing: Exchange of the use of the element.

64. Collaboration: Working together on the use of the element.

65. Partnership: Joint efforts in the use of the element.

66. Consortium: A group of organizations working together on the use of the element.

67. Network: A system of interconnected organizations.

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Virtual Lab 7 Chemistry Answer Key

Jessica J Manson



Virtual Lab 7 Chemistry Answer Key:

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