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Chapter 2

Short Answer Problems

- The LMTD factor is required for heat exchangers when the two streams do not exhibit pure counter-current flow. Most exchangers such as S&T and other arrangements fall into this category.
- This statement is not generally true. In fact, condensers often have high heat transfer coefficients on the condensing side. One exception is for partial condensers where the majority of the condenser-side of the exchanger comprises of non-condensing gas – in this case the heat transfer coefficient on the condensing side may be limiting.
- Turbulent flow on both sides of the heat exchanger see Table 2.9

a. Shell-side
$$h$$
 Ress h Re

4. Laminar flow on both sides of the heat exchanger - - see Table 2.9

a. Shell-side
$$h$$
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- All else considered, the heat transfer coefficient for turbulent flow will be (much) higher than for laminar flow. For this reason when possible flow should be in the turbulent regime.
- Refer to Section 2.2.1.7 three reasons for placing a fluid on the tube side of a S&T exchanger

2-If the fluid is corrosive

2-If the fluid causes severe fouling or scaling

2-If the fluid is at a much higher pressure than the other fluid

2-If the fluid has a much higher film coefficient

2-If the material in contact with the fluid must be an expensive alloy

Solution Manual Chemical Process

John Krenos, Joseph Potenza, Loretta
Jones, Lynn Koplitz, Thomas Spence

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Solutions Manual, Chemical Process Safety, Fundamentals with Applications [by] Daniel A. Crowl [and] Joseph F. Louvar Daniel F. Goerke, Joseph F. Louvar, 1990 Chemical Process Control George Stephanopoulos, 1984 Principles of Chemical Engineering Processes - Solutions Manual Nayef Ghasem, Redhouane Henda, 2008-09-26 Solutions Manual to Accompany Applied Mathematics and Modeling for Chemical Engineers Richard G. Rice, Duong D. Do, James E. Maneval, 2023-05-09 This book is a Solutions Manual to accompany Applied Mathematics and Modeling for Chemical Engineers Third Edition There are many examples provided as homework in the original text and the solution manual provides detailed solutions of many of these problems that are in the parent book Applied Mathematics and Modeling Solutions Manual for Analysis, Synthesis, and Design of Chemical Processes for Chemical Engineers Third Edition Jessica W. Castillo, Richard Turton, Richard C. Bailie, Wallace Whiting, Joseph A. Shaeiwitz, 2012-09-14 Accompany Basic Principles and Calculations in Chemical Engineering David Mautner Himmelblau, James B. Riggs, 2004-01 Chemical Principles Study Guide/Solutions Manual John Krenos, Joseph Potenza, Carl Hoeger, 2007-01-18 Written for general chemistry courses Chemical Principles helps students develop chemical insight by showing the connection between chemical principles and their applications **Exploring Chemical Analysis Solutions Manual Daniel C.** Harris, 2004-04-30 Exploring Chemical Analysis teaches students how to understand analytical results and how to use quantitative manipulations preparing them for the problems they will encounter Student Solutions Manual for Exploring Chemical Analysis Daniel C. Harris, 2008-04-18 Exploring Chemical Analysis teaches students how to understand analytical results and how to use quantitative manipulations preparing them for the problems they will encounter Solutions Manual For Chemical Engineering Thermodynamics Y. V. C. Rao, 1998 This book is a very useful reference that contains worked out solutions for all the exercise problems in the book Chemical Engineering Thermodynamics by the same author Step by step solutions to all exercise problems are provided and solutions are explained with detailed and extensive illustrations It will come in handy for all teachers and users of Chemical Engineering Thermodynamics Solution Manual for Analysis, Synthesis, and Design of Chemical Processes, 2009 **Student Solutions Manual for Physical Chemistry** C. A. Trapp, Peter Atkins, Julio de Paula, 2009-12-18 With its modern emphasis on the molecular view of physical chemistry its wealth of contemporary applications vivid full color presentation and dynamic new media tools the thoroughly revised new edition is again the most modern most effective full length textbook available for the physical chemistry classroom Available in Split Volumes For maximum flexibility in your physical chemistry course this text is now offered as a traditional text or in two volumes Volume 1 Thermodynamics and Kinetics ISBN 1 4292 3127 0 Volume 2 Quantum Chemistry Spectroscopy and Statistical Thermodynamics ISBN 1 4292 3126 2 Chemical Principles Student's Study Guide & Solutions Manual John Krenos, Joseph Potenza, Loretta Jones, Lynn Koplitz, Thomas Spence, 2004-03-19 This combination manual is designed to help

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