

Understanding Solubility

What is the quantity of solute in a saturated solution at a certain temperature?

Activity 1

Prepare two solid solvents: a potassium dichromate (K₂Cr₂O₇) solution for 100 ml water and copper sulphate (CuSO₄) solution for 100 ml water. Add solute to the solvents until it is completely dissolved. In separate beakers, quantity of solute added to saturate a solution at a certain temperature, which then should have some statements that statements will easily observed through practical in group. Report all possible values of solute dissolved in the system.

Observation: In the above, copper sulphate solution, the solute is dissolved in water. In the potassium dichromate solution, the solute is dissolved in water. The solute is dissolved in water.

Table 1: – Solubility of Solutes

The following table shows the solubility of solutes in water at different temperatures. The solubility of solutes in water is given in g/100 ml. The solubility of solutes in water is given in g/100 ml. The solubility of solutes in water is given in g/100 ml. The solubility of solutes in water is given in g/100 ml.

Solute	Experiment 1 in 100 g water		Experiment 2 in 100 g water		Experiment 3 in 100 g water	
	Amount of solute dissolved (g/100 ml)	Amount of solute dissolved (g/100 ml)	Amount of solute dissolved (g/100 ml)	Amount of solute dissolved (g/100 ml)	Amount of solute dissolved (g/100 ml)	Amount of solute dissolved (g/100 ml)
1	10.0	10.0	10.0	10.0	10.0	10.0
2	10.0	10.0	10.0	10.0	10.0	10.0
3	10.0	10.0	10.0	10.0	10.0	10.0
4	10.0	10.0	10.0	10.0	10.0	10.0
5	10.0	10.0	10.0	10.0	10.0	10.0
6	10.0	10.0	10.0	10.0	10.0	10.0
7	10.0	10.0	10.0	10.0	10.0	10.0
8	10.0	10.0	10.0	10.0	10.0	10.0
9	10.0	10.0	10.0	10.0	10.0	10.0
10	10.0	10.0	10.0	10.0	10.0	10.0
11	10.0	10.0	10.0	10.0	10.0	10.0
12	10.0	10.0	10.0	10.0	10.0	10.0
13	10.0	10.0	10.0	10.0	10.0	10.0
14	10.0	10.0	10.0	10.0	10.0	10.0
15	10.0	10.0	10.0	10.0	10.0	10.0

1. Solubility of solutes in water is given in g/100 ml. The solubility of solutes in water is given in g/100 ml.

2. Solubility of solutes in water is given in g/100 ml. The solubility of solutes in water is given in g/100 ml.

3. Solubility of solutes in water is given in g/100 ml. The solubility of solutes in water is given in g/100 ml.

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