## Solubility and Temperature

## Answer Key

Vocabulary: concentration, dissolve, homogeneous mixture, solubility, solubility curve, solute, solution, solvent

#### Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

[Note: The purpose of these questions is to activate prior knowledge and get students thinking. Students are not expected to know the answers to the Prior Knowledge Questions.]

What happens when you stir a spoonful of sugar into hot water?

The sugar dissolves into the water.

When sugar or another substance is dissolved in water, it disappears from view and forms a homogeneous mixture with the water, also called a solution.

If you can't see the sugar, how can you tell that it is there?

The water tastes sweet and may appear slightly cloudy. [You can boil or evaporate the water to see solid sugar again.]

Does sugar dissolve more easily in hot water or cold water? Hot water

#### Gizmo Warm-up

A solution generally consists of two parts, a **solute** that is dissolved and a **solvent** that the solute is dissolved into. For example, sugar is a solute that is dissolved into the solvent water. In the Solubility and Temperature Gizmo<sup>TM</sup>, you will study how temperature affects how much solute will dissolve in a solution.

To begin, check that Potassium nitrate is selected and the Temp, of the water is 20 °C. Click OK.

- 1. In this solution, what is the solute? Potassium nitrate
- And 1 g And 10 g And 20 g
- What is the solvent? Water
- Click Add 10 g to mix 10 g of potassium nitrate into the water.
  - A. Did all of the potassium nitrate dissolve? Yes
  - B. How can you tell?

All of the potassium nitrate disappeared as it was stirred into the water.

# **Solubility And Temperature Answer Key**

David Sang, Peter Ellis, Derek McMonagle

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