

SOLUBILITY CURVE WORKSHEET KEY

Use your solubility curve graphs provided to answer the following questions.

1. What are the customary units of solubility on solubility curves? Degrees Celsius and grams of solute/100g of water
2. Define solubility. A measure of how much solute can dissolve in a given amount of solvent.
3. According to the graph, the solubility of any substance changes as temperature changes.
4. List the substances whose solubility decreases as temperature increases. NaCl and CaSO_4
5. Which substance is least affected by temperature changes? NaCl
6. How many grams of ammonium chloride (NH_4Cl) at 50°C ? 50g
7. NaCl and KCl have the same solubility at approximately 75°C .
8. Which compound is least soluble in water at 10°C ? KClO_3
9. How many grams of KNO_3 can be dissolved at 50°C ? 80g
10. Are the following solutions unsaturated, saturated, or supersaturated?
 - a. 40g of NaNO_3 in 100 g of water at 30°C . saturated
 - b. 60g of KClO_3 in 100 g of water at 90°C . supersaturated
11. How many grams of sodium chloride, NaCl are required to saturate 100 grams of water at 100°C ? 40g
12. How many grams of NaNO_3 are required to saturate 100 grams of water at 75°C ? 140g
13. How many grams of KI will saturate water at 20°C ? 30g
14. At what temperature would 25g of potassium chlorate (KClO_3) dissolve? 35°C
15. At what temperature would 60g of NH_4Cl dissolve? 35°C
16. 80-g NaNO_3 is prepared at 30°C .

Solubility Packet Answers

**United States. National Bureau of
Standards**

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the text box.

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Nieuwenhuis,2005-07-14 This book constitutes the refereed proceedings of the 20th International Conference on Automated Deduction CADE 20 held in Tallinn Estonia in July 2005 The 25 revised full papers and 5 system descriptions presented were carefully reviewed and selected from 78 submissions All current aspects of automated deduction are addressed ranging from theoretical and methodological issues to presentation and evaluation of theorem provers and logical reasoning systems

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