

## \* Solid state \*

- \* **Solid** → have definite mass, volume, shape  
 → short intermolecular distance  
 → strong intermolecular force  
 → incompressible & rigid  
 → particle have fixed position & only oscillate about their position.

### \* solid have 2 types

#### crystalline

→ The solid which have definite order of arrangement of particle  
 ex. NaCl, Quartz

#### Amorphous

→ The solid have irregular arrangement of particle  
 ex. glass, rubber, plastic, Quartz glass

| properties           | crystalline                                          | Amorphous.                                    |
|----------------------|------------------------------------------------------|-----------------------------------------------|
| shape                | definite geometric shape                             | irregular shape.                              |
| Melting              | Melt at sharp temp.                                  | Gradually soften over range of temp.          |
| cleavage             | when cut newly generated surface are plain & smooth  | when cut new generated surface are irregular. |
| heat of fusion       | definite heat of fusion                              | not have definite heat of fusion.             |
| Anisotropy           | Anisotropic (properties change in differ. direction) | isotropic (properties same in all direction)  |
| Nature               | True solid                                           | Pseudo solid or super cooled liquid           |
| order of arrangement | Long range order                                     | only short range order.                       |

# Solid State Hsc In

**Singh Kamal & Singh S.P.**



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