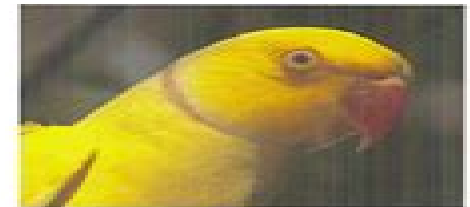


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oxidation $\text{Co} \rightarrow \text{Co}^{2+} + 2\text{e}^{-}$ reduction $\text{Fe}^{2+} + 2\text{e}^{-} \rightarrow \text{Fe}$ 24 3 $\text{Ag} + \text{Ni}^{3+} \rightarrow \text{Ni} + 3\text{Ag}^{+}$ oxidation reduction 25 $\text{Cu}^{2+} + \text{Pb} \rightarrow \text{Pb}^{2+} + \text{Cu}$ oxidation reduction 26 $\text{O}_2 + 2\text{H}_2 \rightarrow 2\text{H}_2\text{O}$

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 $5\text{Co} + \text{I}_2 + 5\text{Co}_2\text{I}_2 + 5\text{O}_2 \rightarrow 5\text{CoI}_2 + 5\text{Co}_2\text{O}_2$

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