

CDX-GT21W/GT210/ GT260/GT260S

SERVICE MANUAL

Ver. 1.2 2007.03



US Model
Canadian Model
CDX-GT21W/GT210
E Model
CDX-GT260/GT260S
Chinese Model
CDX-GT260

- The tuner and CD sections have no adjustments.

AUDIO POWER SPECIFICATIONS (US, CND MODEL)
POWER OUTPUT AND TOTAL HARMONIC DISTORTION
23 watts per channel minimum continuous average power into
4 ohms, 4 channels driven from 20 Hz to 20 kHz with no more
than 5% total harmonic distortion.

Model Name Using Similar Mechanism	CDX-GT11W/GT110/ GT160/GT160S
CD Drive Mechanism Type	MG-101TA-168/Q
Optical Pick-up Name	DAX-25A

SPECIFICATIONS

CD player section

Signal-to-noise ratio 95 dB
Frequency response 19 – 20,000 Hz
Wow and flutter Below measurable limit

Tuner section

FM

Tuning range 87.5 – 107.9 MHz (US, CND model)
87.5 – 108 MHz (at 50 kHz step) (E, CH model)
87.5 – 107.9 MHz (at 200 kHz step) (E, CH model)
50 kHz/200 kHz switchable (E, CH model)
FM tuning interval
Antenna terminal External antenna connector
Intermediate frequency 10.7 MHz/450 kHz
Usable sensitivity 9 dB
Selectivity 75 dB at 400 kHz
Signal-to-noise ratio 67 dB (stereo), 69 dB (mono)
Harmonic distortion at 1 kHz 0.5% (stereo), 0.3% (mono)
Separation 35 dB at 1 kHz
Frequency response 39 – 15,000 Hz

AM/MW

Tuning range 530 – 1,710 kHz (US, CND model)
531 – 1,602 kHz (at 9 kHz step) (E, CH model)
530 – 1,710 kHz (at 10 kHz step) (E, CH model)
9 kHz/10 kHz switchable (E, CH model)
AM tuning interval
Antenna terminal External antenna connector
Intermediate frequency 10.7 MHz/450 kHz
Sensitivity 30 μ V

SW (E, CH model)

Tuning range SW1: 2,940 – 7,735 kHz
SW2: 9,500 – 18,135 kHz
(except for 10,140 – 11,575 kHz)
Antenna (aerial) terminal External antenna (aerial) connector
Intermediate frequency 10.7 MHz/450 kHz
Sensitivity 30 μ V

– Continued on next page –

FM/AM COMPACT DISC PLAYER

CDX-GT21W/GT210

FM/MW/SW COMPACT DISC PLAYER

CDX-GT260/GT260S

Sony Cdx Gt260mp Manual

Mark Steyvers



Sony Cdx Gt260mp Manual:

Sony Cdx Gt260mp Manual Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Sony Cdx Gt260mp Manual**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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web describe redox reactions in terms of electron transfer recall that oxidation involves loss of electrons and reduction involves gain of electrons represent simple redox reactions using half equations identify oxidising and reducing agents for simple redox reactions understand the reactivity of metals and metal displacement reaction

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web write the oxidation and reduction reactions for each redox reaction the first one is done for you $23 \text{ Fe} + \text{Co} \rightarrow \text{Co} + 2 \text{ Fe}$
oxidation $\text{Co} \rightarrow \text{Co}^{2+} + 2 \text{e}^{-}$ reduction $\text{Fe}^{2+} + 2 \text{e}^{-} \rightarrow \text{Fe}$ $24 \text{ 3 Ag} + \text{Ni} \rightarrow \text{Ni}^{3+} + 3 \text{ Ag}$ oxidation reduction $25 \text{ Cu} + 2 \text{ Pb} \rightarrow \text{Pb}^{2+} + \text{Cu}$ oxidation reduction 26 O_2

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5co i2 5co2 i 2 o 5 5 c o i 2 5 c o 2

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web chemical reactions in which there is a transfer of electrons from one substance to another are known as oxidation reduction reactions or redox reactions in this unit you will examine the oxidation reduction process and use the oxidation state and oxidation number concepts to identify redox reactions and to keep track of electrons

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web jun 28 2010 answers i determine what is oxidized and what in each reaction identify the oxidizing agent and the reducing agent also $2\text{sr} + \text{o}_2 \rightarrow 2\text{sro}$ sr0 to sr2 oxidized reducing agent o0 to o2 reduced ox ag $2\text{li} + \text{s} \rightarrow \text{li}_2\text{s}$ li0 to li1 oxidized red ag s_0 to s_2 reduced ox ag $3\text{cs} + \text{br}_2$